



Pergamon

TETRAHEDRON

Tetrahedron 54 (1998) 3647–3692

Tetrahedron report number 446

Recent Developments in General Methodologies for the Synthesis of Linear Triquinanes[§]

Vishwakarma Singh* and Beena Thomas

Department of Chemistry, Indian Institute of Technology, Bombay-400 076, India

Received 8 December 1997

Contents

1. Introduction	3648
2. General Methods Leading to Triquinanes in a Single Step	3649
2.1. 1,3-Diyl trapping	3649
2.2. Olefin to arene <i>meta</i> -photocycloaddition	3652
2.3. Photo-thermal metathesis reaction	3655
2.4. Radical initiated polyolefinic cyclizations	3656
2.5. Pauson–Khand reaction	3660
2.6. Squarate ester-1,3,5,7-octatetraene-polyquinane cascade	3664
2.7. Tandem oxy-Cope-transannular ring closure route to polyquinanes	3667
2.8. Photocycloaddition-fragmentation approach	3668
2.9. Zinca-ene–allene reaction	3669
2.10. Palladium catalyzed cyclopentanation of alkenes	3671
2.11. One pot synthesis via conjugate addition of enolate-carbanions	3672
2.12. Palladium catalyzed cyclization of enynes	3672
2.13. Diels–Alder cycloaddition-oxa-di- π -methane rearrangement	3676
3. General Methods Leading to Triquinanes via Diquinanes	3679
3.1. Intramolecular Diels–Alder approach	3679
3.2. Palladium promoted syntheses	3682
3.3. Radical cyclization	3684
4. Summary and Outlook	3685

[§]Dedicated to Prof. E. Osawa and Dr. V. V. Kane on the occasion of their 62nd and 65th birthday, respectively.

1. INTRODUCTION

Polyquinane is a generic name given to carbocyclic frames composed of fused five membered rings and constitutes an important class of sesquiterpenoids. Since the discovery of polyquinane natural products they have generated a world wide interest among organic chemists due to their unique and fascinating molecular architecture and promising biological activity.¹ Among polyquinanes, the natural products having triquinane framework are more abundant. At present nearly eighty such natural products are known in the literature and they are frequently encountered among plants,² marine³ and microbial sources.⁴ Triquinane natural products containing all the three types of C-11 tricyclopentanoid skeleta, **1** (linearly fused five membered rings), **2** (angularly fused five membered rings) and **3** (five membered rings fused in propellane fashion), as the core ring systems are known (Fig.-1).^{5,6}

The class of linearly fused tricyclopentanoids is further divided depending upon the mode of fusion of the third cyclopentane ring. The two isomers **4** and **5** are termed as *cis:anti:cis* and *cis:syn:cis*, respectively. Although, these two tricyclic hydrocarbons have not been characterized yet, their ΔH_f° values have been calculated by Osawa and coworkers.⁷ It has been shown that **4** is only marginally more stable than the hindered folded form **5**, which nevertheless forms the basic building block of aesthetically pleasing polyhedra such as peristyrene **10**, dodecahedrane⁸ **11** (Fig.-2) and a variety of molecular hosts.⁹

Of the two stereoisomeric C₁₁ linear triquinanes, *cis:anti:cis* isomer **4** has received relatively greater attention because it constitutes the basic carbocyclic framework of a large number of naturally occurring triquinane sesquiterpenoids such as hirsutic acid **6**, coriolin **7**, capnellene **8** and hirsutene **9** (Fig.-1).

The intense interest in the chemistry of polycyclopentanoids is partly due to novel and intricate carbocyclic network of natural triquinanes and also because of diverse biological properties exhibited by some members. For example, hirsutic acid **6** has antibiotic properties while coriolin **7** shows antibacterial and antitumor activities.⁶ Capnellene **8** and its congeners have been suggested to act as chemical defense agents to inhibit growth of micro organisms and to prevent larval settlement.¹⁰ As a consequence, the last decade has witnessed a flurry of activity in the design and development of synthetic routes to cyclopentanoids.¹¹⁻¹³ The search for methods for the efficient and rapid synthesis of polyquinanes is continuing unabated.¹⁴⁻¹⁶

The literature on the synthesis of polyquinanes is vast and expanding and, comprehensive reviews^{11a,b} on the subject are available covering the literature up to 1988.^{11c} In this report, we concentrate on general methodologies leading to the linear *cis:anti:cis* triquinanes to reflect the recent proliferation of work in this area.

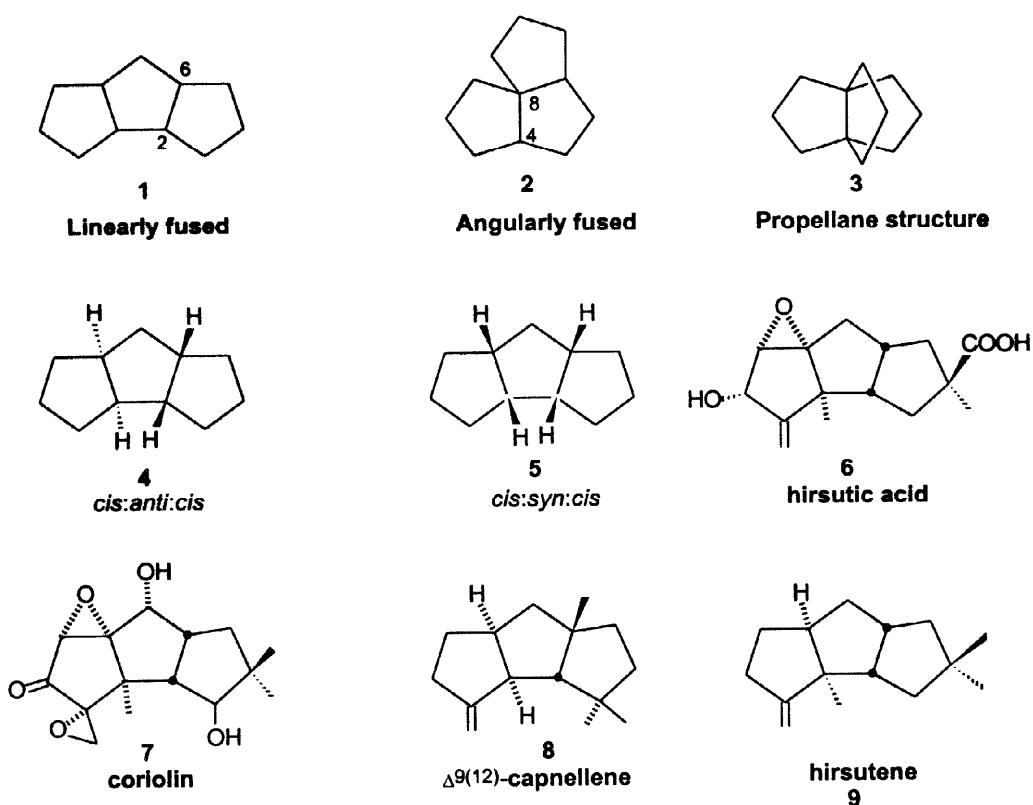


Fig.-1

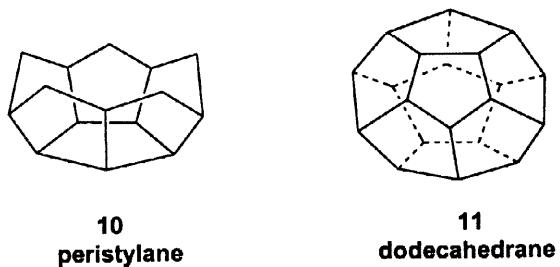


Fig.-2

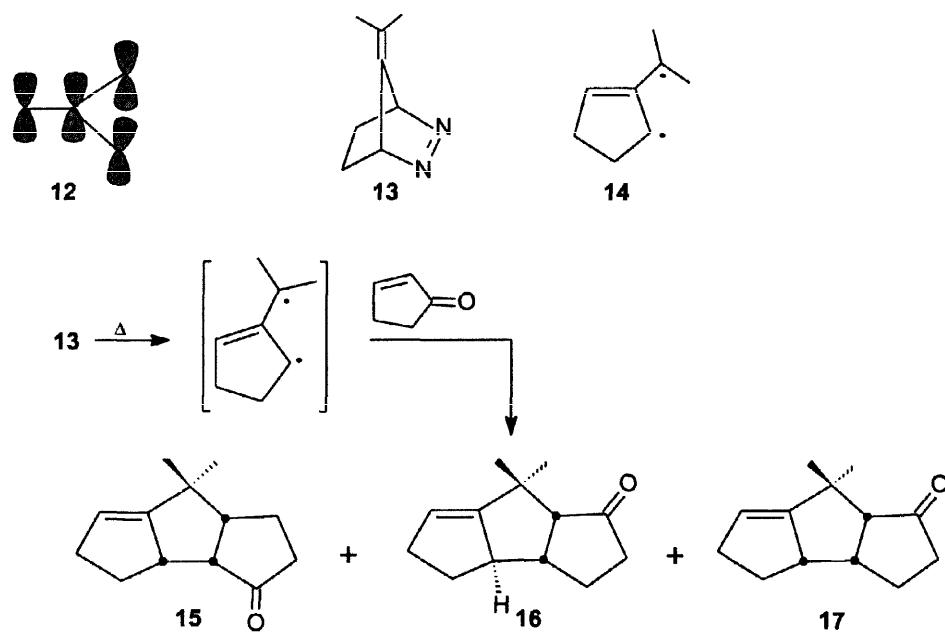
2. GENERAL METHODS LEADING TO TRIQUINANES IN A SINGLE STEP

In this section, those methods which generate the tricyclopentanoid framework directly in a single step from appropriate precursors have been presented.

2.1. 1,3-DIYL TRAPPING

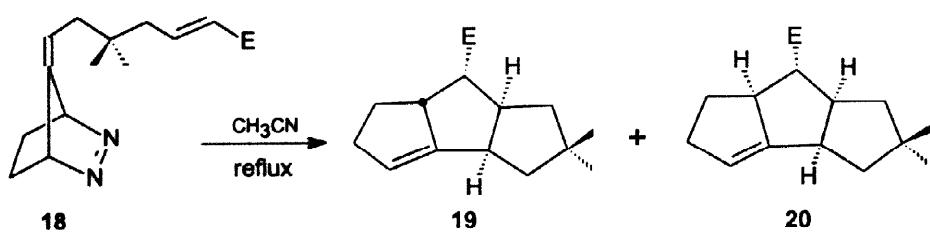
Little and coworkers¹⁷⁻²⁰ developed a new general method for synthesis of triquinane framework. The method involves *in situ* generation and trapping of a highly reactive diradical (diyl) of type 14²¹ which is

analogous to trimethylenemethane²² (TMM) 12. Their initial approach to tricyclopentanoids employed pyrolysis of the azo compound 13 to diyl 14 and its interception with cyclopentenone which led to various products (15-17) (Scheme-1). Analysis of the initial results, particularly the regio- and stereoselectivity of the intermolecular cycloaddition, led them to reason that an intramolecular version of the above reaction might prove to be more selective. Indeed, the pyrolysis of the azo compound 18 furnished two tricyclopentanoids 19 and 20 in a ratio of 9:1 and in very good yield (85%) (Scheme-2). The mechanistic details and the factors affecting the regio- and stereoselectivity of these reactions have been now discussed in detail.²⁰ After exploratory studies on the intramolecular diyl trapping, Little and his associates demonstrated the potential of the above method through synthesis of hirsutene,¹⁸ coriolin¹⁹ and capnellene.²⁰ The syntheses of coriolin and capnellene are briefly described here.

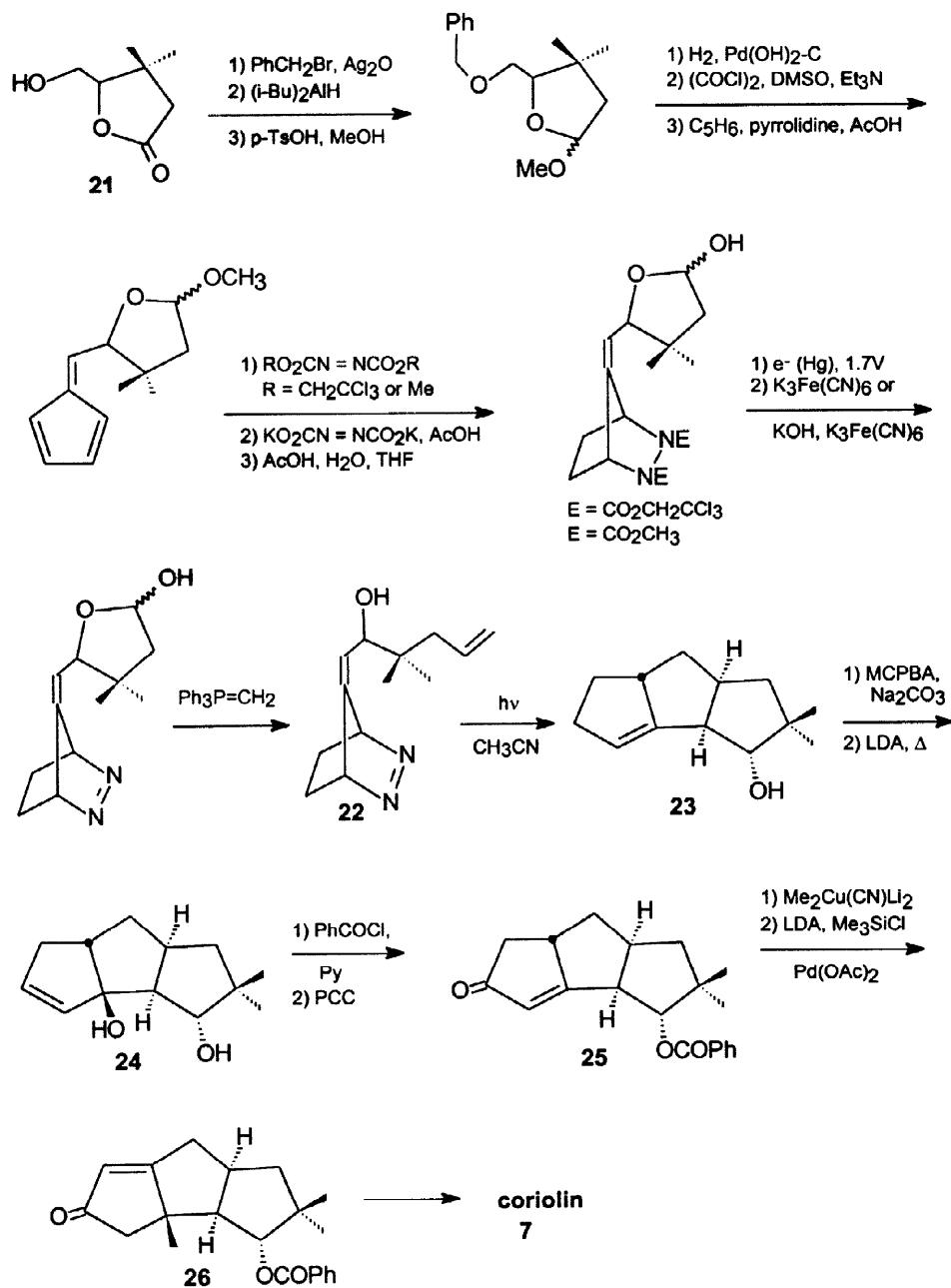


Scheme-1

In an application of the above methodology towards coriolin synthesis, the required azo compound 22 was prepared by a series of reactions from dihydro-5-(hydroxymethyl)-4,4-dimethyl-2(3H)-furanone 21 as shown in the Scheme-3. The facile conversion of 22 to the linear triquinane 23 was effected under photolytic conditions. The alcohol 23 was transformed into the enediol 24 by epoxidation with MCPBA, followed by Rickborn-Crandall^{19c,d} ring opening initiated by LDA. Protection of the secondary alcoholic group and oxidation with PCC gave the enone 25. Conjugate addition and generation of the double bond using LDA furnished 26. The enone 26 has previously been converted into coriolin by Koreeda *et al.*²³

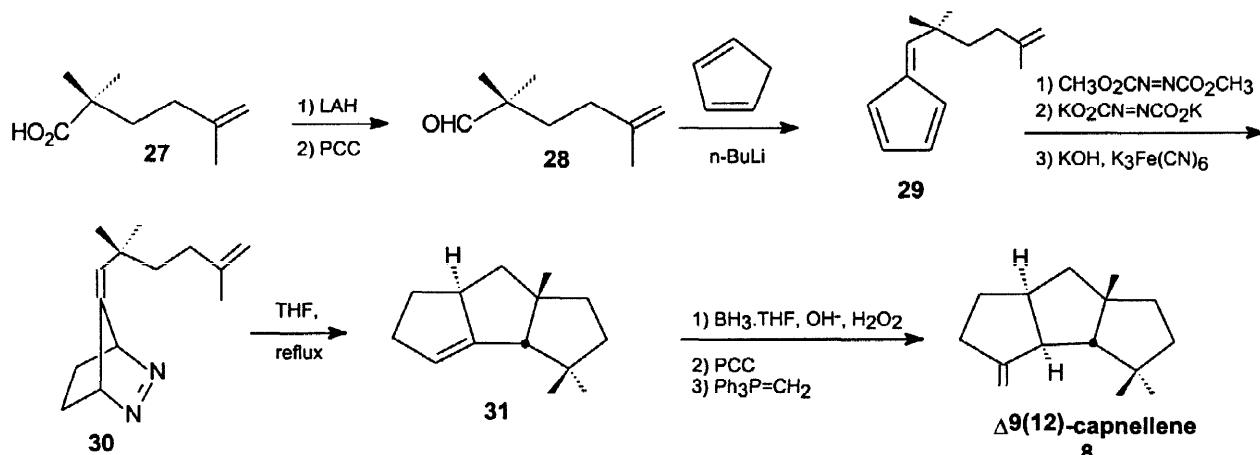


Scheme-2



Scheme-3

The synthesis of $\Delta^{9(12)}$ -capnellene following the above methodology is delineated in the Scheme-4. The carboxylic acid **27** was first reduced to the corresponding alcohol, which was then oxidized with PCC to **28**. Subsequent fulvene formation using cyclopentadienyllithium, set the stage for a Diels-Alder reaction between fulvene **29** and dimethyl azodicarboxylate. The resulting dicarbamate was transformed efficiently into **30** which was converted to the triquinane **31** on heating under reflux in tetrahydrofuran. The hydrocarbon **31** on the hydroboration-oxidation sequence followed by PCC oxidation and Wittig olefination gave the desired natural product, $\Delta^{9(12)}$ -capnellene **8**.



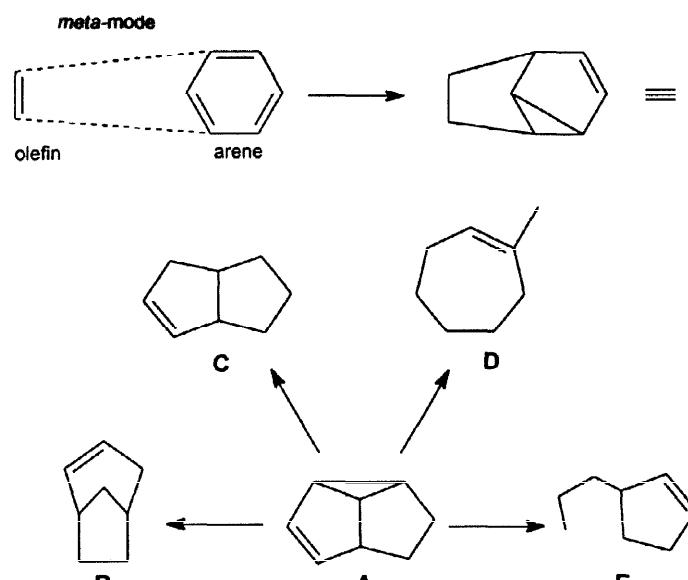
Scheme-4

The 1,3-diyi trapping method offers a simple and convenient route to triquinanes. The triquinane skeleton is built-up in one step having desired *cis:anti:cis* stereochemistry, though the preparation of the starting azo compound requires several steps. For example, the triquinane skeleton for the synthesis of hirsutene¹⁸ and coriolin is generated after nine and twelve steps respectively. Moreover, in the case of synthesis of hirsutene, the key step resulted in the formation of isomers.

2.2. OLEFIN TO ARENE META-PHOTOCYCLOADDITION

The photocycloaddition of alkenes to arenes provides a potent tool for the synthesis of many novel and otherwise not accessible products. Among the three theoretically possible addition modes,²⁴ the *ortho*-, *meta*- and *para*-cycloadditions, the *meta*- type or 1,3-cycloaddition, especially in intramolecular fashion,²⁵ has proven to be of great value in natural product synthesis as demonstrated by Wender.²⁶ Three mechanisms have been suggested for the course of *meta*-cycloaddition: (a) interception of the isomerized biradicaloid aromatic by the olefin; (b) formation of an exciplex²⁵ of charge-transfer character,^{27,28} or (c) a concerted process. The mechanistic data are in best accord with an exciplex mechanism for the singlet - state²⁷⁻²⁹ which explains satisfactorily the regio- and stereoselective course of such reactions. The carbon

skeleton of the product, a tricyclo[3.3.0.0^{2,8}]octene [A], is a versatile synthetic intermediate, which allows entry into different structural units of high utility²⁶ (Scheme-5).

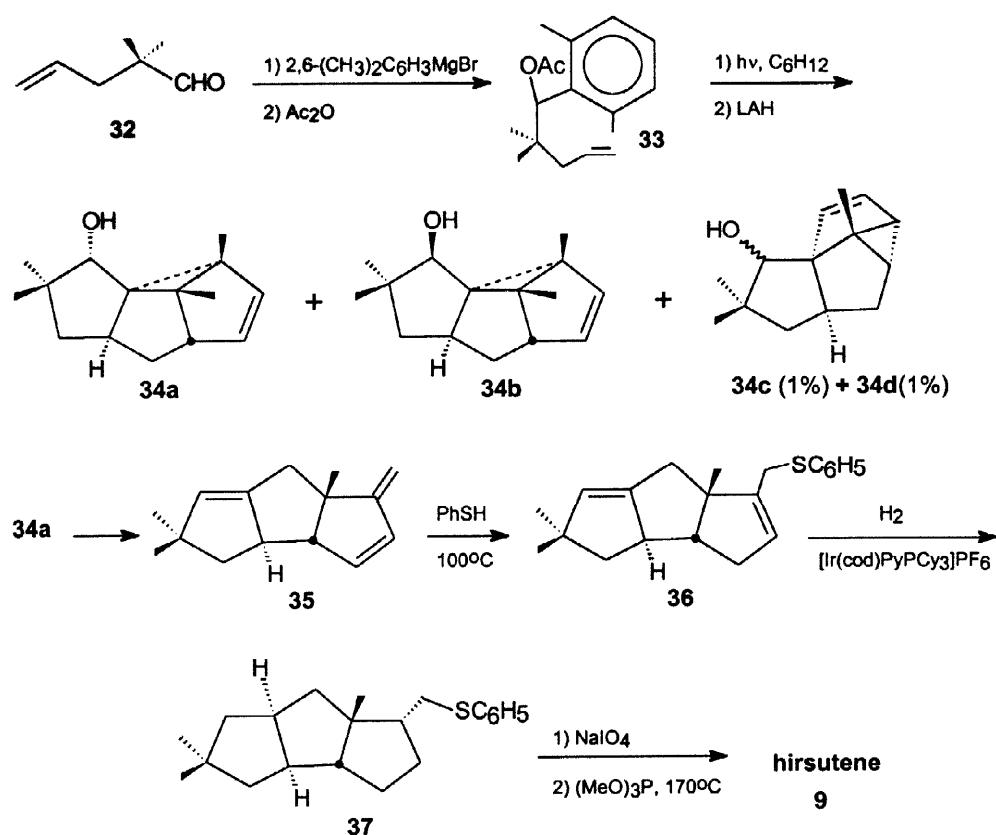


Scheme-5

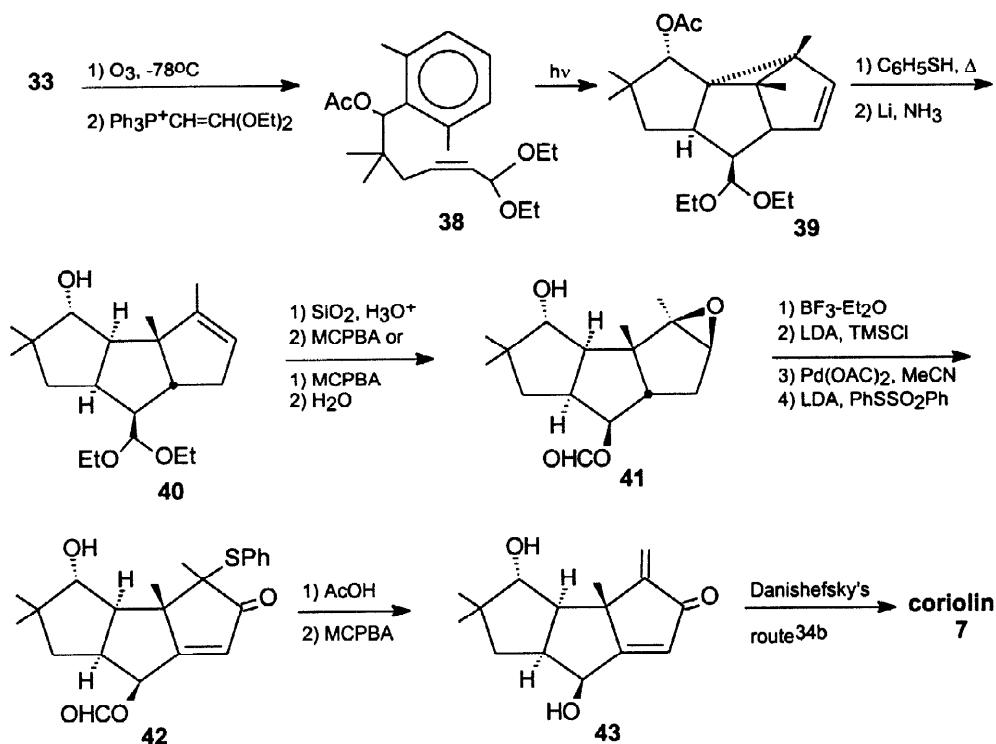
The method was first applied in the synthesis of (\pm)-cedrene,³⁰ a natural product with dicyclopentanoid moiety (A \rightarrow B, transformation) and subsequently in the synthesis of angular triquinanes (\pm)-isocomene,³¹ silphenene³² and silphiperfol-6-ene³³ in order to prove its generality. Intramolecular *meta*-cycloaddition route was also exploited for the synthesis of linear tricyclopentanoids, hirsutene 9 and coriolin 7 as described below.

In the hirsutene synthesis, Wender and Howbert³⁴ started with 2,2-dimethyl-4-pentenal 32 as depicted in the Scheme-6. The requisite starting material 33 for the intramolecular *meta*-photoaddition was prepared by addition of the aryl-Grignard reagent of 2,6-dimethyl bromobenzene to 32 followed by protection of the secondary alcohol function. The photocycloaddition of 33 in intramolecular fashion offered 34a in 23% yield along with other isomers (34b-d). Dehydrative acid-catalyzed rearrangement of 34a led to 35. Free radical 1,4-addition of thiophenol on 35 gave the compound 36 which was hydrogenated to give 37. Oxidation of 37 to sulfoxide followed by elimination under standard conditions regenerated the exocyclic olefin to give hirsutene.

In the coriolin synthesis,^{34a} 33 was transformed into 38 which upon photocycloaddition gave the tricyclic acetate 39. The overall synthesis of coriolin from the triquinane 39 is shown in Scheme-7. Thus, the acetate 39 was hydrolysed to give the alcohol 40 which was converted into the enone 42 via the epoxyformate 41. The intermediate 42 was transformed into the key precursor 43 as shown in the Scheme-7.



Scheme-6

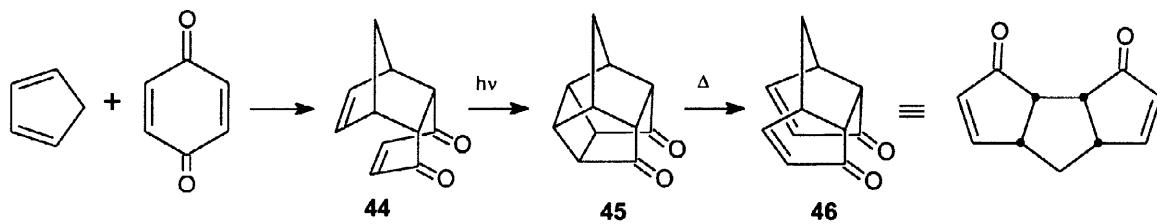


Scheme-7

Although the tricyclopentanoid framework is rapidly achieved by this method, the key photochemical step gives a mixture of the cycloaddition products depending upon the mode of cycloaddition and such products have to be separated. Moreover, a multi-step sequence is required to generate the desired starting material.

2.3. PHOTO-THERMAL METATHESIS REACTION

Mehta and coworkers³⁵ have developed a novel and versatile route for the synthesis of *cis:syn:cis* tricyclo[6.3.0.0^{2,6}]undecane (tricyclopentanoid) framework via a novel photo-thermal metathetic sequence. This method involves a photochemical $\pi^2s + \pi^2s$ cycloaddition in *endo* adducts of type **44** and a regiospecific thermal fragmentation of the cyclobutane ring of the resulting product **45** to give *cis:syn:cis* triquinane **46**. The strategy is depicted below (Scheme-8). The generality of this sequence is demonstrated by synthesis of a large number of *cis:syn:cis* triquinanes bearing a variety of substituents. It is interesting to note that *cis:syn:cis* triquinane framework thus generated was isomerized to *cis:anti:cis* framework and hence provided an opportunity to develop syntheses of many natural products. Application of this methodology to the synthesis of capnellene **8**, coriolin **7** and hirsutene **9** is described below.

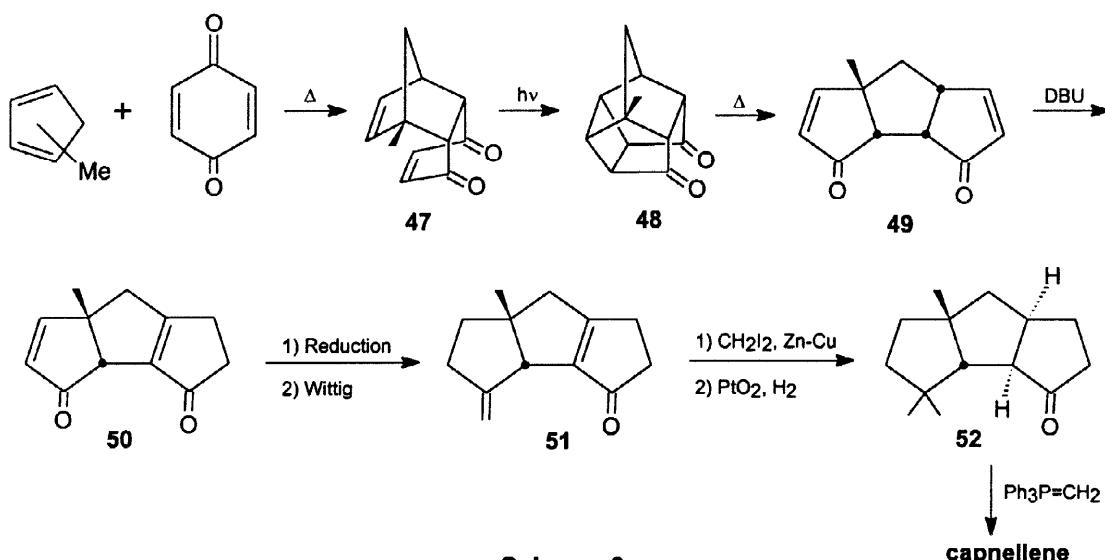


Scheme-8

The $(\pm)\Delta^{9(12)}$ -capnellene synthesis (Scheme-9) of Mehta and coworkers³⁶ employed Diels–Alder reaction of isomeric methylcyclopentadienes and *p*-benzoquinone from which the desired tricyclic *endo*-adduct **47** was obtained. Its intramolecular photocycloaddition to **48** followed by thermal fragmentation gave **49**. It was treated with base to obtain the diene-dione **50**. Selective catalytic hydrogenation of **50** followed by olefination gave **51**. The treatment of **51** with CH_2I_2 in the presence of Zn–Cu couple gave the spirocyclopropane derivative which was hydrogenated to give the intermediate **52**. Since the ketone **52** has already been transformed into capnellene,³⁷ the formal synthesis was complete.

Mehta's group have used a common intermediate for the synthesis of coriolin and hirsutene. In the coriolin synthesis³⁸ (Scheme-10), the tricyclic system **53** was obtained from thermal Diels–Alder reaction between cyclopentadiene and 2,5-dimethyl benzoquinone. The adduct **53** was subjected to intramolecular [2+2] photocycloaddition which furnished the cage dione **54**. Flash vacuum pyrolysis of **54** delivered the compound **55** which on thermal equilibration and hydrogenation afforded the *cis:anti:cis* dione **56**. The dione

56 was alkylated regioselectively to yield dimethyl dione **57**. Chemoselective addition of MeMgI to the less hindered carbonyl group of **57** followed by dehydration gave **58** which was reduced stereoselectively with Li-NH_3 to obtain thermodynamically more stable alcohol **59**. The alcohol **59** was elaborated to **61** via the ketoalcohol **60** as shown in Scheme-10. Phenyl selenylation-selenoxide elimination of **61** gave the precursor **62** which had previously been converted into (\pm) -coriolin **7**.³⁹

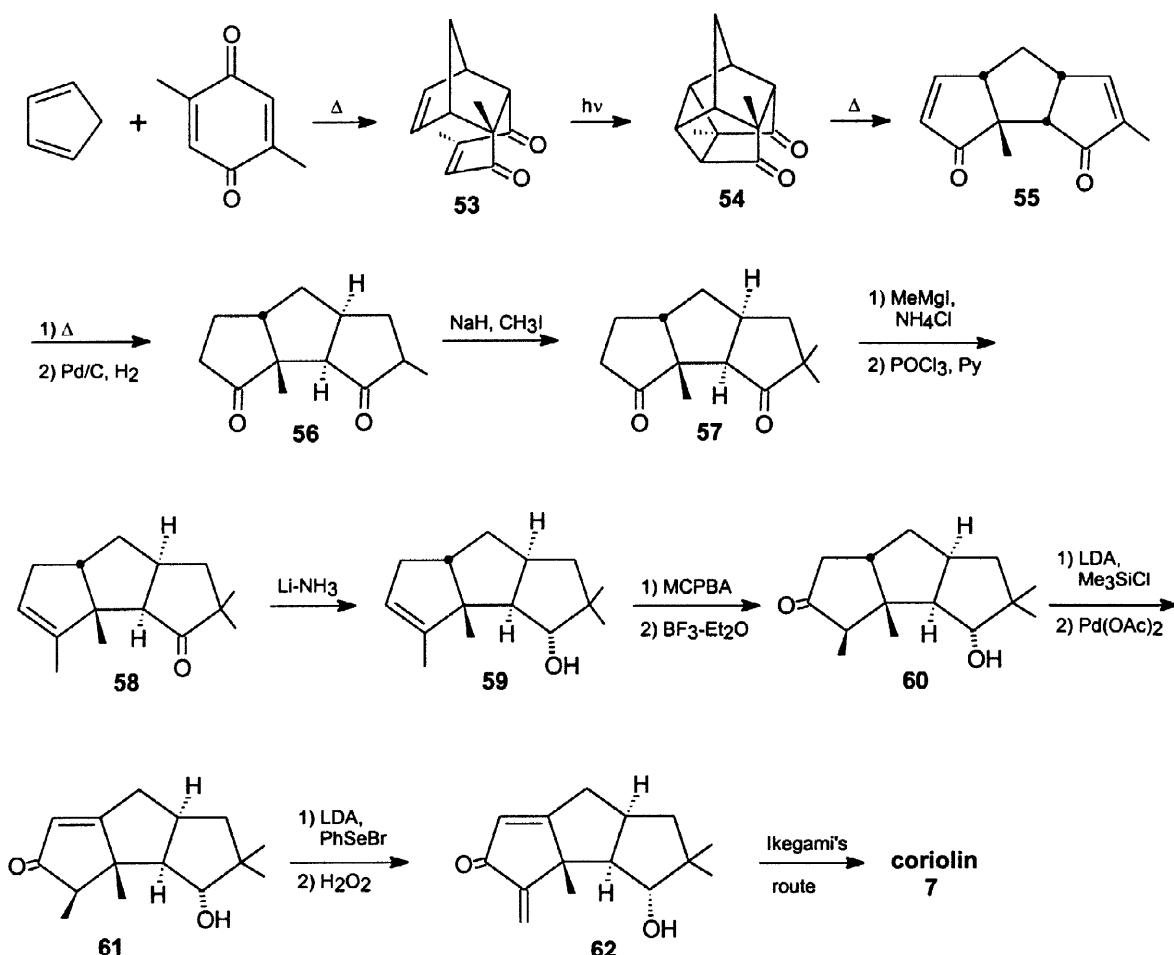


Mehta's hirsutene synthesis (Scheme-11)⁴⁰ follows the same route upto compound **56**. The compound **56** was selectively methylated with potassium *t*-butoxide to furnish **57**. Regioselective Wittig reaction of **57** gave the enone **63**, which was converted to **64**. Transformation of **64** to **65** followed by reduction with tributyltin hydride furnished the hirsutene **9**.

2.4. RADICAL INITIATED POLYOLEFINIC CYCLIZATIONS

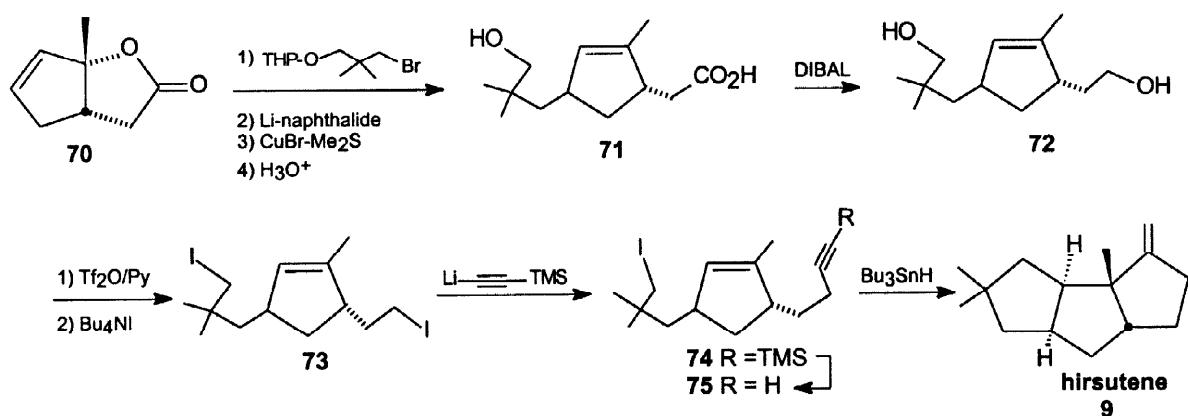
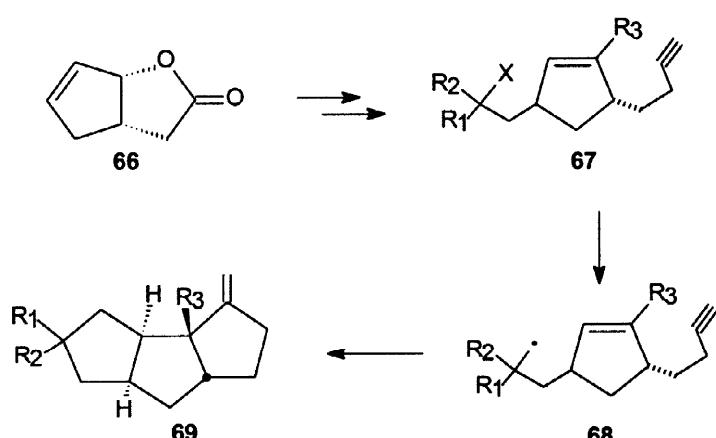
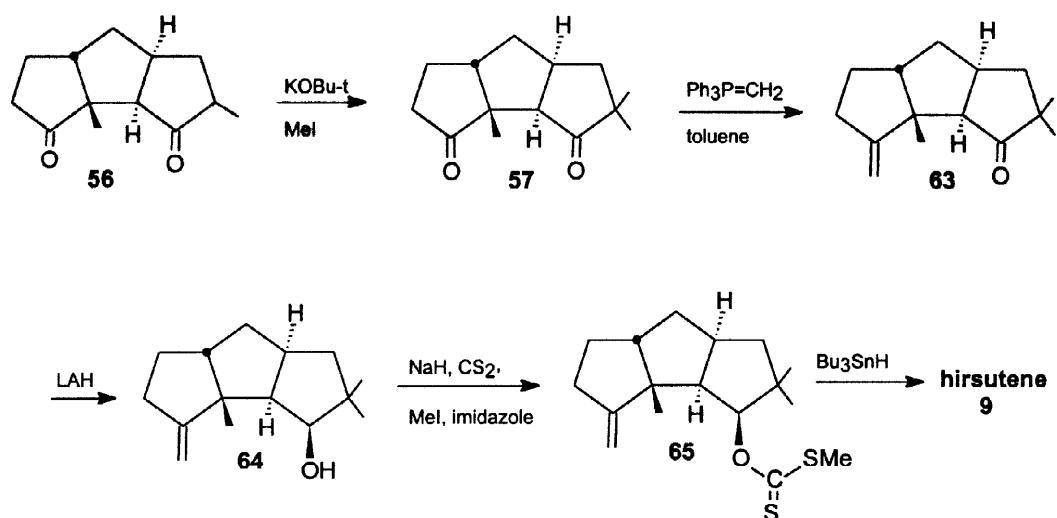
Curran and his associates have developed^{41,42} a novel unified strategy for the synthesis of linear condensed cyclopentanoids employing a tandem hex-5-enyl radical cyclization as the key step.^{43–45} The ability of radical cyclizations to construct multiple five membered rings in a controllable fashion has been featured in syntheses of the three major classes of triquinanes (linear, angular and propellane). The synthesis is executed in three steps: (1) $\text{S}_{\text{N}}2\text{-anti}$ opening of a vinyl lactone **66** to provide a *trans*-disubstituted cyclopentene **67**; (2) rapid elaboration to a cyclization precursor **68**; and (3) tandem radical cyclization leading to **69**. A tandem radical cyclization forms the two outer rings of a triquinane about a central pre-formed cyclopentene ring (Scheme-12). The generality and flexibility of this strategy are quite apparent in that variations in the nature and disposition of the side chains can lead to different triquinanes. The study of the mechanistic data indicates that in most cases these cyclizations are stereoelectronically

controlled, irreversible, proceed regardless of the degree of substitution of starting and product radicals and tolerate a wide variety of pendant functional groups.^{42,43} The synthetic potential of the above method has been illustrated by efficient total syntheses of (\pm)-hirsutene,^{41,42} capnellene⁴⁶, coriolin and hypnophilin.⁴⁷

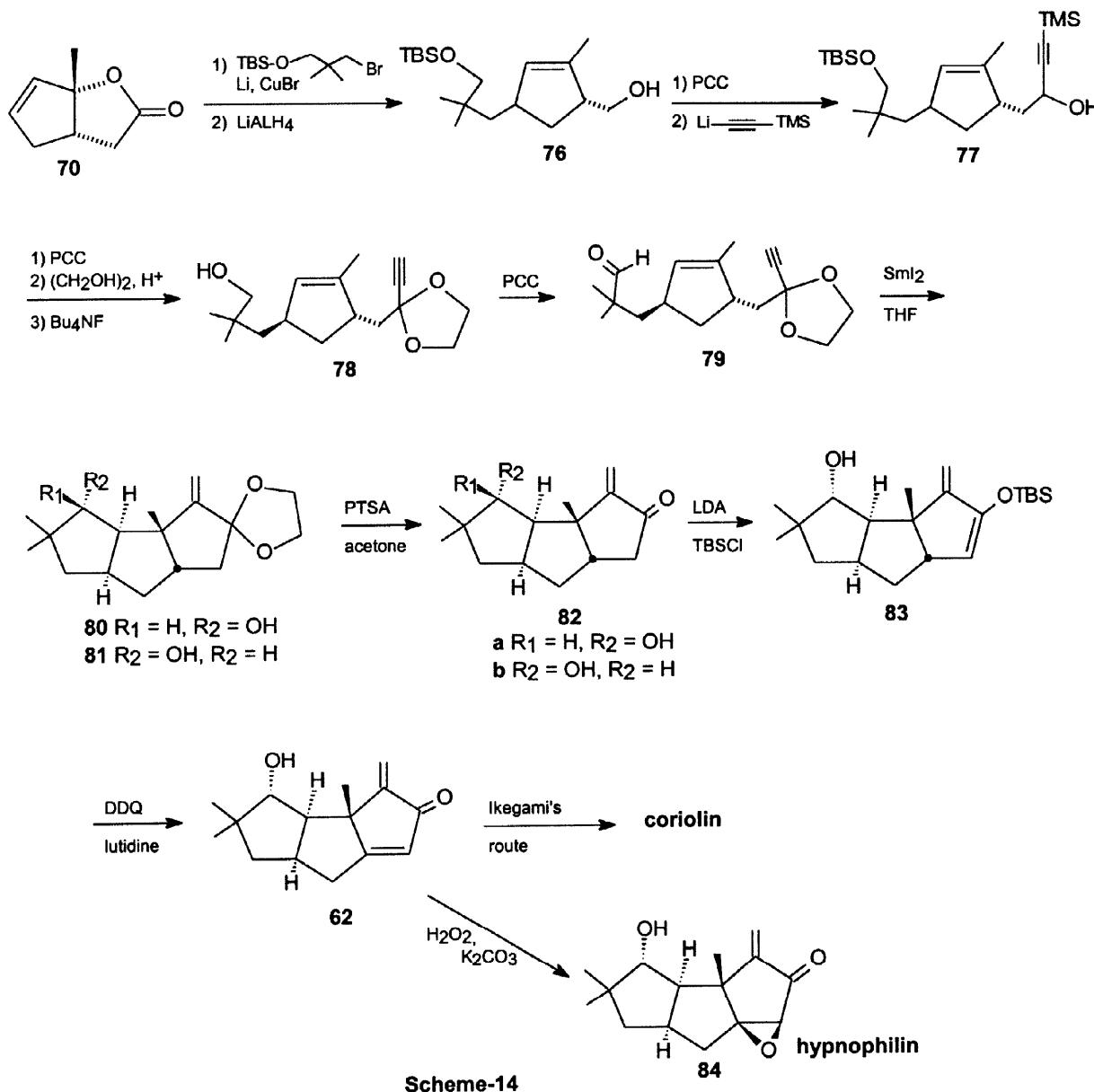


Scheme-10

Towards application of this methodology for the synthesis of hirsutene, the vinyl lactone **70** was first converted into the *trans*-disubstituted hydroxy-acid **71** (Scheme-13). Reduction of **71** with DIBAL gave the diol **72** which was then transformed into the diiodide **73** as shown in the Scheme-13. Treatment of **73** with lithium trimethylsilyl acetylide furnished the silyl acetylene **74** which was desilylated with tetrabutylammonium fluoride to give **75**. Treatment of **75** with tributyltin hydride gave hirsutene in a single step (Scheme-13).

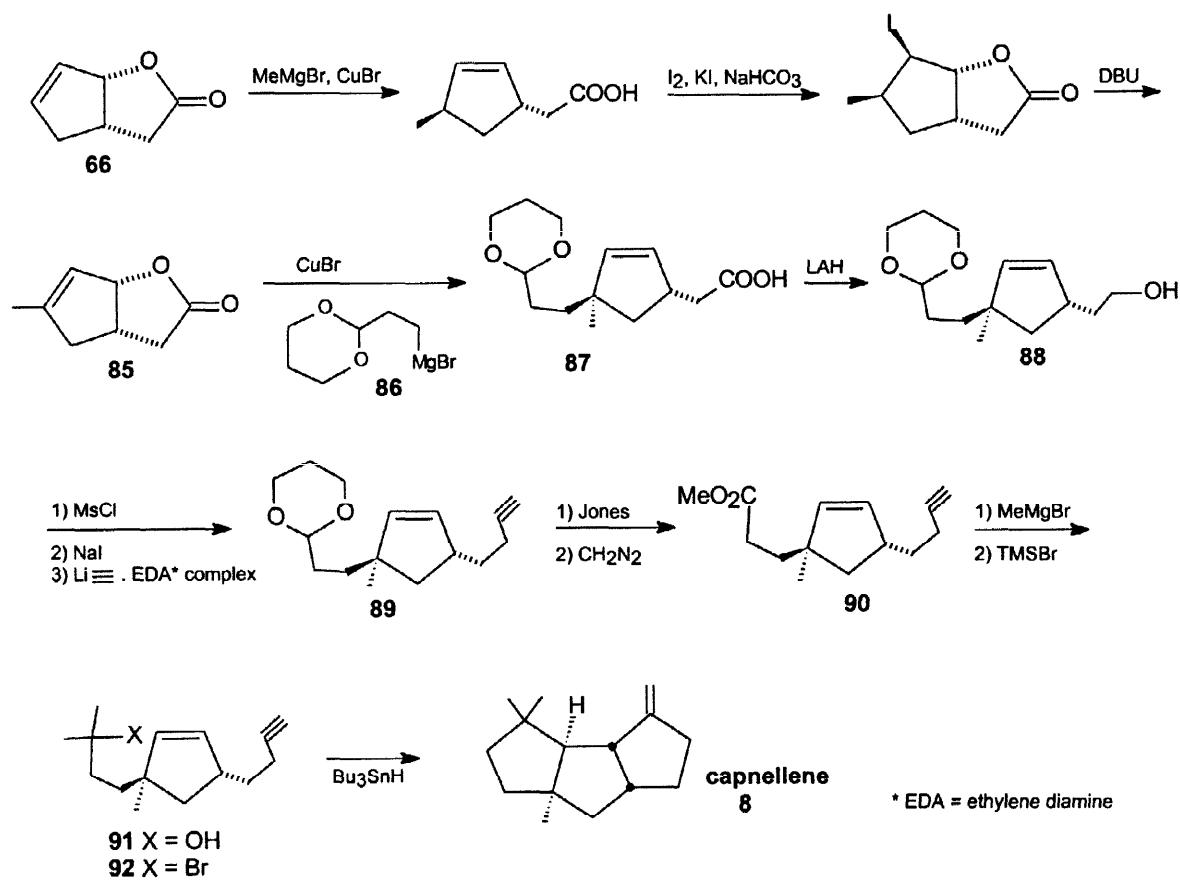


The same lactone **70** was transformed into acetylenic ketone **79** for coriolin synthesis (Scheme-14). Herein, **70** was converted to the *trans*-substituted cyclopentene derivative **76**. Oxidation of **76** followed by a nucleophilic addition with the silyl acetylidyde gave **77**. The compound **77** was then transformed into the protected acetylidyde **78** which upon oxidation gave **79** as shown in Scheme-14. Samarium(II) catalyzed cyclization of **79** furnished stereoisomeric alcohols **80** and **81** which contain most of the structural and stereochemical features of coriolin. Deprotection of **80** and **81** with pTSA gave the enones **82a, b**. **82a** was then converted to the silylenol ether **83**. Introduction of the double bond in **83** gave the intermediate **62**, a precursor to coriolin in an earlier synthesis.^{39b} Epoxidation of **62** with H_2O_2 gave hypnophilin **84** (Scheme-14).⁴⁷



Scheme-14

Synthesis of capnellene has also been completed along similar lines (Scheme-15).⁴⁶ The vinyl lactone **85** was synthesized from the unsubstituted vinyl lactone **66** via an S_N2 -anti opening, iodolactonization followed by a base promoted elimination reaction sequence. The reaction of **85** with the species generated by addition of the Grignard reagent **86** to an equivalent of CuBr/DMS complex, produced the acid **87** which upon reduction with lithium aluminium hydride gave the alcohol **88**. The compound **88** was converted into the enyne **89**. Removal of the protective group and oxidation with Jones reagent followed by treatment of the resulting acid with diazomethane gave the ester **90**. The addition of excess methylmagnesium bromide produced the tertiary alcohol **91** which gave the cyclization precursor **92** on treatment with trimethylsilyl bromide. Cyclization of **92** with tributyltin hydride in presence of AIBN produced $\Delta^{9(12)}$ -capnellene.

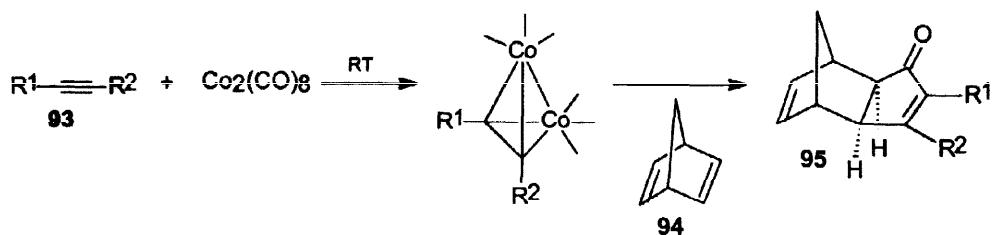


Scheme-15

2.5. PAUSON-KHAND REACTION

The [2+2+1] cycloaddition between an alkyne, alkene and CO (from $\text{Co}_2(\text{CO})_8$) leading to cyclopentenone is called the Pauson–Khand reaction, for example, the formation of **95** by reaction of enyne **93** and norbornadiene (**94**) (Scheme-16).

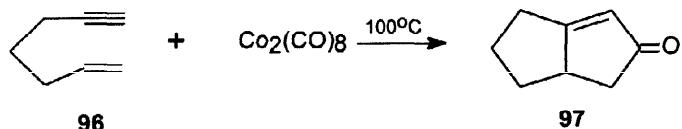
* EDA = ethylene diamine



Scheme-16

This reaction was first reported by Pauson and Khand in 1973⁴⁸ and now constitutes one of the popular approaches to cyclopentenone ring formation. The Pauson–Khand reaction has been well studied⁴⁹ and it has now been established⁵⁰ that: a) the yields of the cyclopentenones are improved if the alkene is strained; b) the reaction is regiospecific with the larger group of the alkyne being accommodated adjacent to the newly inserted CO group; c) the *cis*-exo ring fusion is achieved; d) with respect to the alkene regiochemistry the larger allylic substituent is placed *anti* to the new CO.

Schore reported the first intramolecular version of Pauson–Khand (IMPK) reaction to make the bicyclic system **97**⁵¹ from the enyne **96** (Scheme-17). The first synthetic application of the intramolecular Pauson–Khand reaction was reported by Magnus and coworkers for the synthesis of coriolin⁵² and hirsutic acid⁵³ as described below. The approach involved syntheses of the diquinane precursors which were transformed into triquinanes *via* annulation.

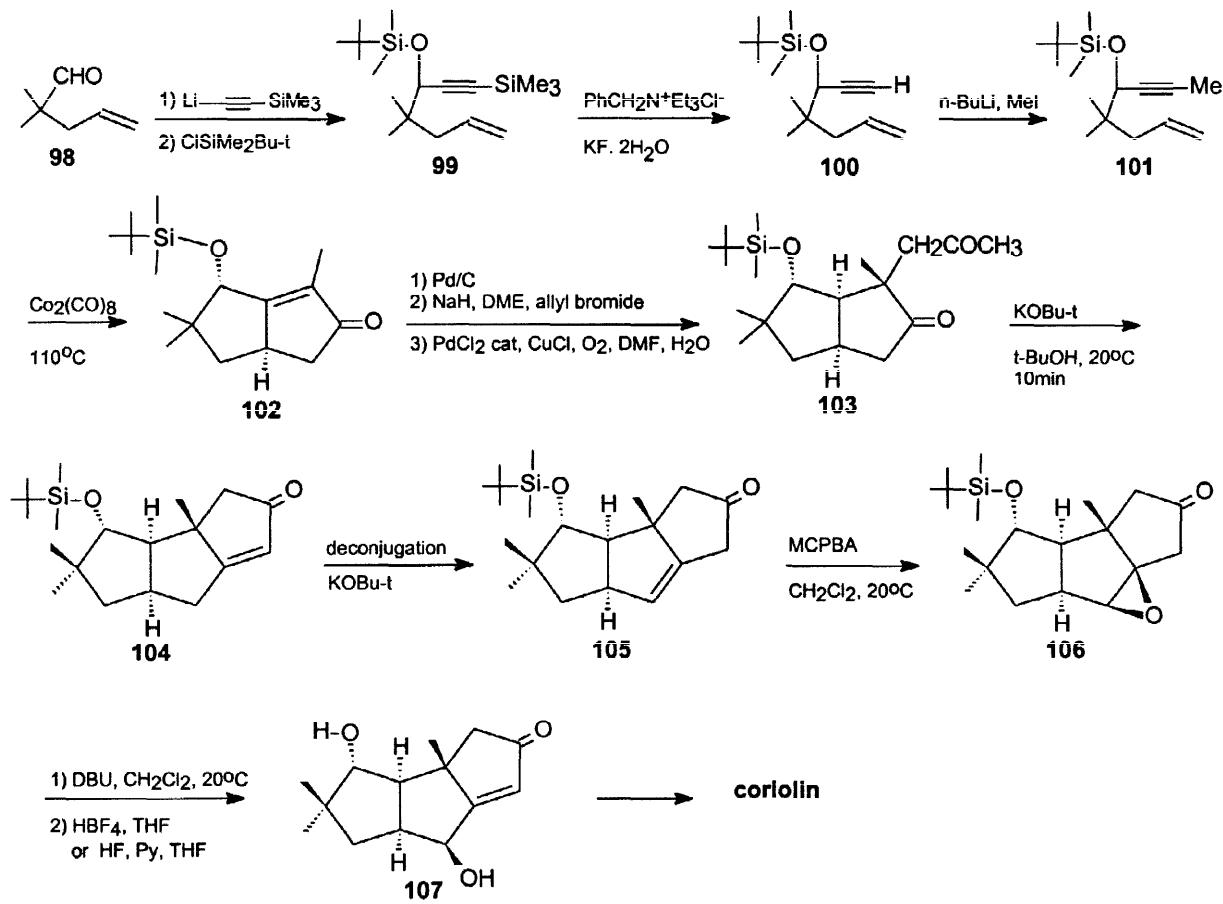


Scheme-17

In the coriolin synthesis, the acyclic aldehyde **98** was converted into **99**. Removal of trimethylsilyl group gave the enyne **100** which was alkylated to furnish the precursor **101**, as shown in Scheme-18. Intramolecular Pauson–Khand reaction of **101** gave the diquinane **102** which upon reduction, allylation and Wacker type oxidation gave the dione **103**. Aldol condensation of **103** furnished the triquinane **104** which was transformed into the enone **105**. Epoxidation of **105** gave the epoxy ketone **106** which was converted into the key precursor **107**. Since the compound **107** has already been converted by Danishefsky^{34b} into coriolin, the formal synthesis was complete.

In a formal synthesis of hirsutic acid Magnus and coworkers prepared the diquinanes **112a** and **113a** which have already been taken to hirsutic acid. Thus, the malonic ester derivative **108** was transformed into the required alicyclic precursor **109** which through the Pauson–Khand reaction gave a mixture of the diquinane **110a**, with the correct stereochemistry and the undesired diquinane **110b**. Treatment of the

mixture of **110a, b** with $\text{MsOH} \cdot 2\text{H}_2\text{O}$ again gave a mixture of acids **111a, b**. The subsequent hydrogenation of this mixture gave **112** which upon esterification with CH_2N_2 yielded a mixture of esters **113a, b**. The epimer **111b** was re-equilibrated to **111a** with *p*-toluene sulphonic acid and therefore **111a** was synthesized from **110** in 90% yield (Scheme-19).

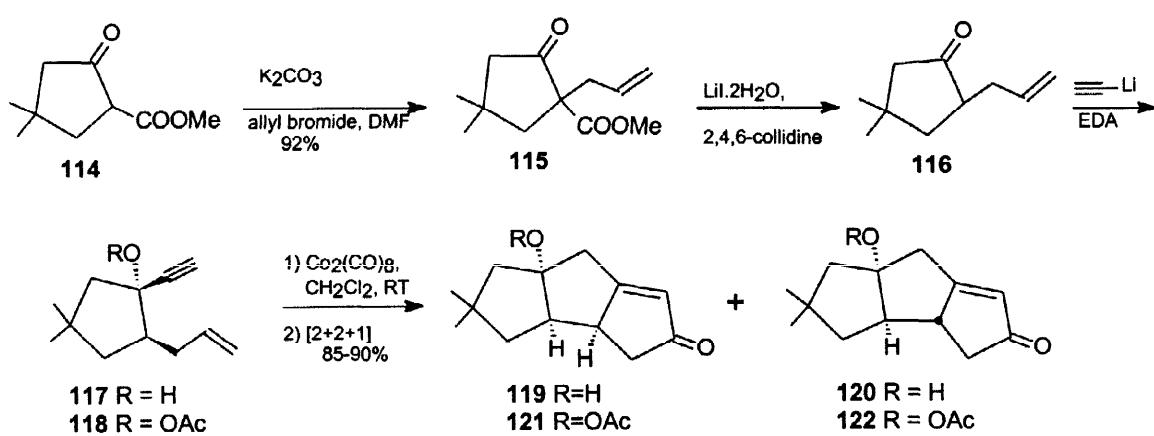
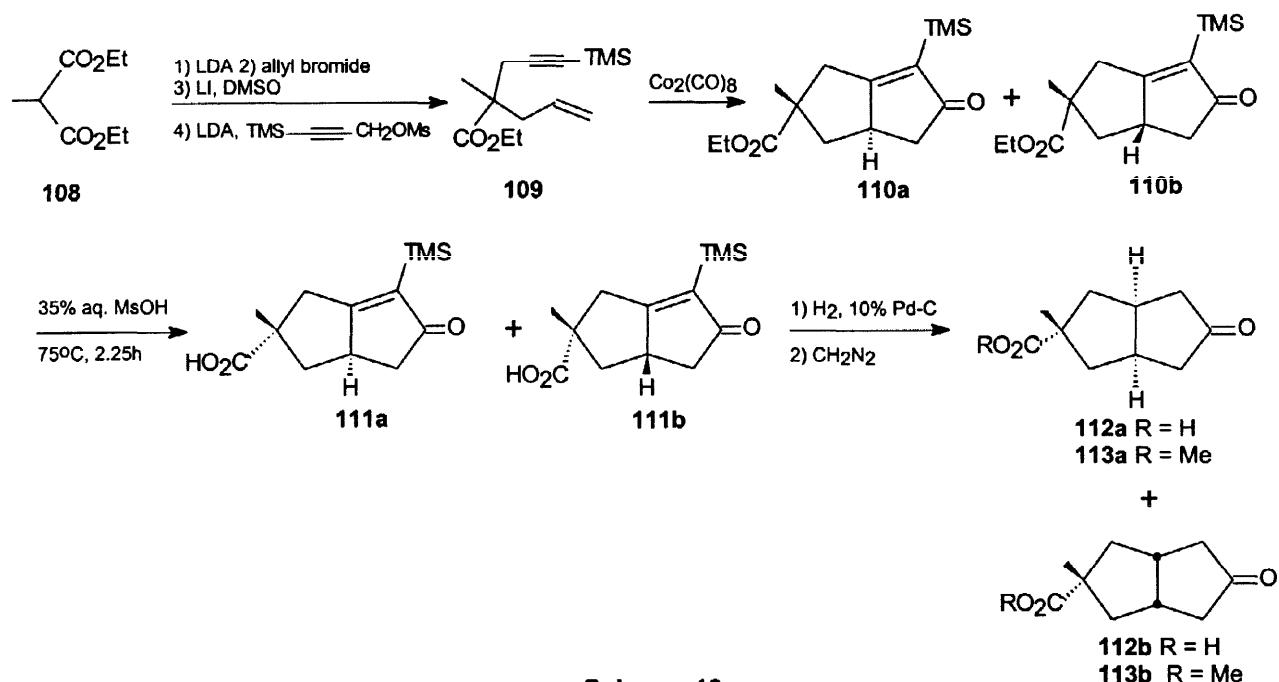


Recently, Veretenov⁵⁴ and associates have been able to assemble the linear triquinane framework *via* a pathway described as “monocyclic precursor → triquinane ring formation” and this method has been employed for the synthesis of triquinanes **119-122** (Scheme-20).⁵⁵ The crucial step in the scheme is the generation of the correctly oriented enyne precursor **117** and its Pauson–Khand reaction. Thus, the *gem*-dimethylated carbomethoxy derivative of cyclopentanone **114** was allylated to **115** and subsequently demethoxycarboxylated to give **116**. The ethynylation of **116** proceeded with moderate efficiency to the required isomer **117** with *cis*-oriented allyl and ethynyl residues being formed with high stereoselectivity. The alcohol **117** and its acetate **118** were converted to dicobaltohexacarbonyl complexes upon treatment with $\text{Co}_2(\text{CO})_8$. The IMPK reaction was carried out on **117** and **118** by thermolysis in solution or on surface of

silicagel or on oxidative initiation of the reaction with *N*-MMO to give 119, 120 and 121, 122, respectively. The stereoselectivity of the reaction is still to be optimized for general applicability.

Hua⁵⁶ used the Pauson-Khand product, for the first time, to synthesize angularly fused triquinane skeletons of optically active pentalenene and racemic pentalenolactone-(*E*)-methyl ester. In another approach Billington and coworkers⁵⁷ carried out the IMPK reaction on an alicyclic enyne precursor, allyl propargyl ether, and applied it to the synthesis of an O-heterocyclic natural product aucubigenone.

The shortness and flexibility coupled with a high degree of stereocontrol are the highlights of the IMPK reaction.



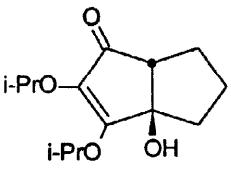
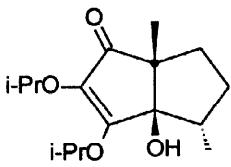
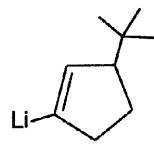
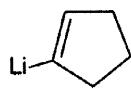
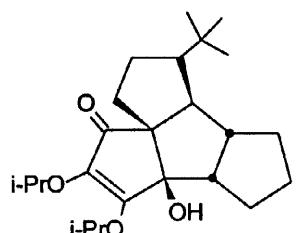
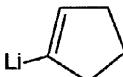
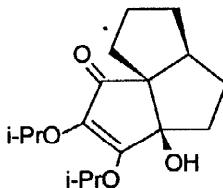
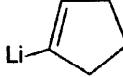
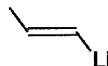
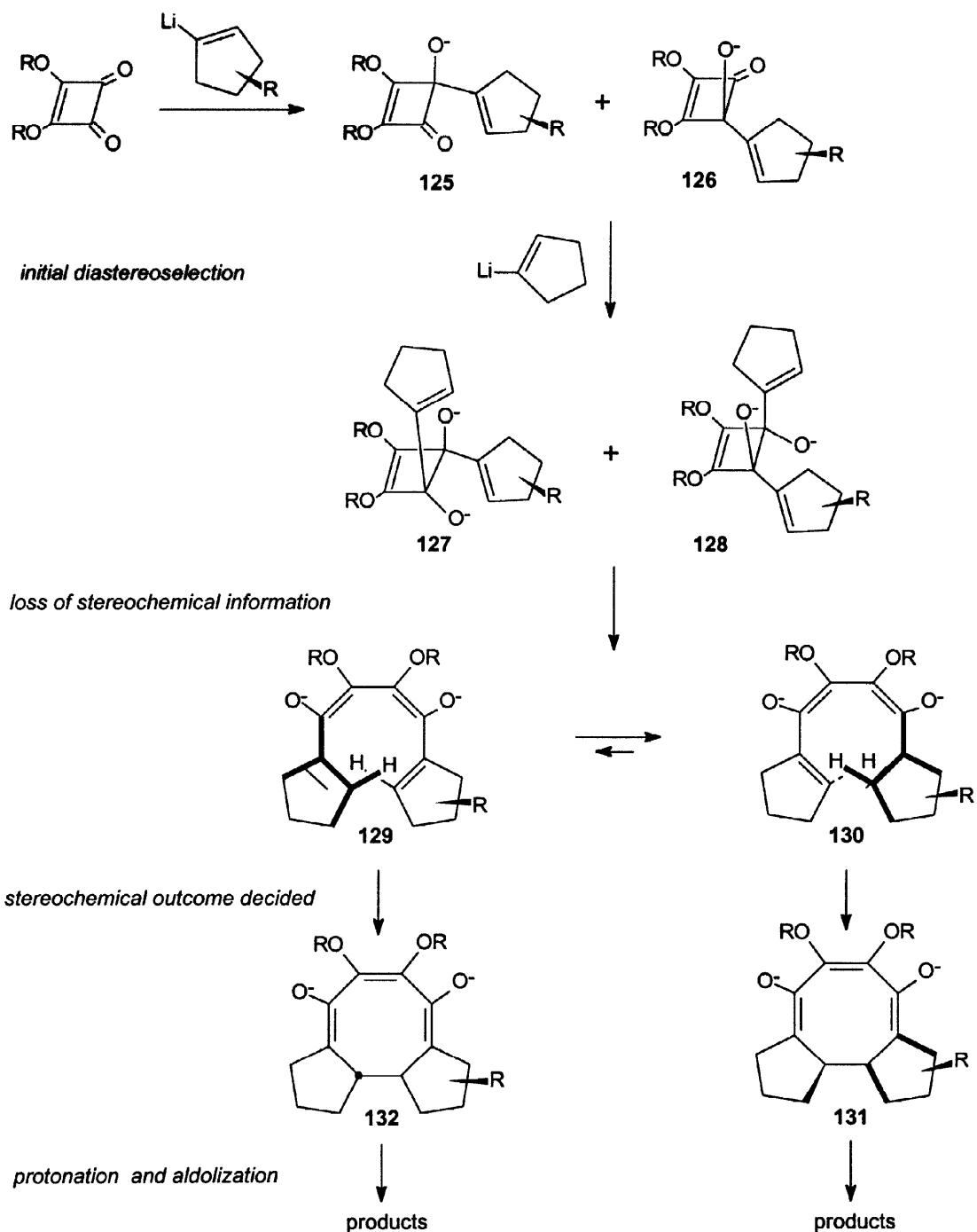
	First anion	Second anion	Products
1.	$\text{LiCH}=\text{CH}_2$ (3eq.)		
2.			
3.			
4.			
5.			 

Table 1



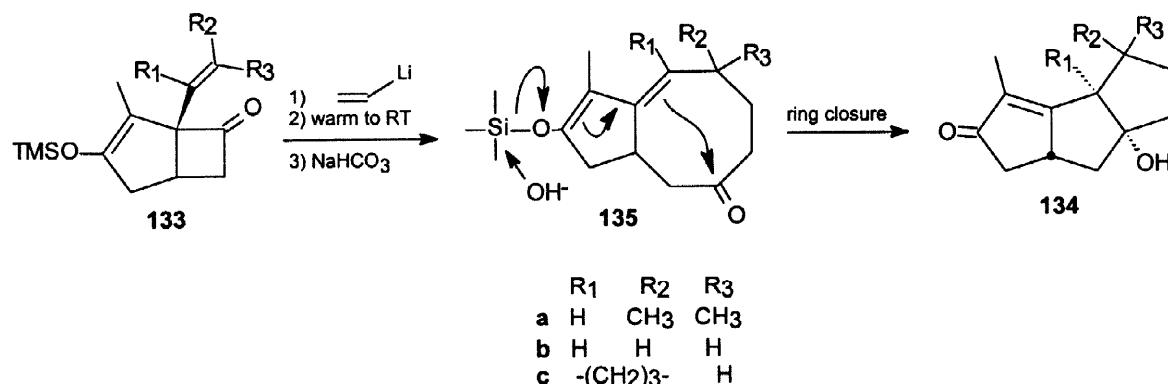
Scheme-22

2.7. TANDEM OXY-COPE-TRANSANNULAR RING CLOSURE ROUTE TO POLYQUINANES

Tandem reactions have become an important tool for construction of complex and functionalized carbocycles in as few steps as possible, and an example of this is the tandem oxy-Cope-transannular ring closure. Santora and Moore⁶³ have described preliminary results of the application of this sequence leading to the synthesis of highly functionalized polycyclopentanoids. The initial step is the addition of an appropriate vinylolithium reagent to bicyclo[3.2.0]heptenones **133**, and oxy-Cope rearrangement of the resulting species leading to the formation of the intermediate such as **135**, which upon transannular cyclization gives highly functionalized polyquinanes (Scheme-23). This approach like the previous one also utilizes the high reactivity of the cyclobutene dimethyl squareate ester to generate the bicycloheptenone precursors **133**. The bicyclic precursors employed in this approach are synthesized from cyclobutadiendione.

The choice of the bicyclo[3.2.0]heptenones and substituted vinylolithium reagent determined the substitution pattern in the products as shown (Table-2, Equations 1-4). Thus, this tandem reaction provides a regiospecific route to highly substituted triquinanes. Substitution patterns are realized by stereoselective alkylation of the enolate intermediate prior to transannular ring closure, for example, addition of vinylolithium to bicycloheptenone followed by warming and quenching of the resulting enolate with methyl iodide afforded the triquinane **134a** (Table-2, Eq. 1). The addition of 2-lithiopropene to **133b** gave the diastereomer **134b**. Further, additional functionality may be introduced into the products by use of substituted vinyl anions.

The synthesis of the linear tetraquinane (Eq. 3) and its angular analogue (Eq. 4) from **133b** and **133c** respectively, shows the generality of the sequence in construction of higher order polyquinanes.



Scheme-23

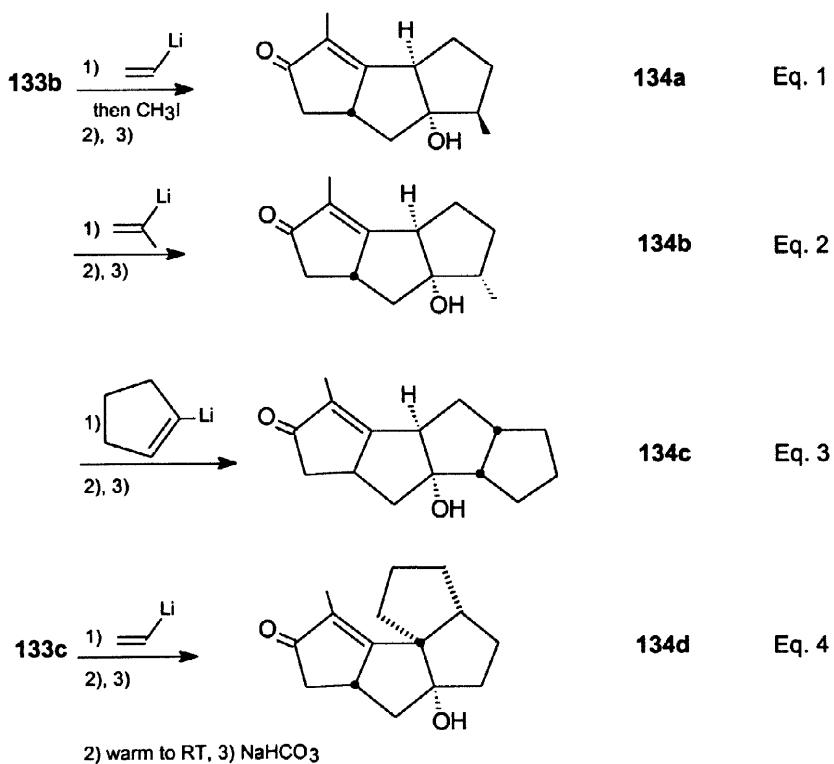
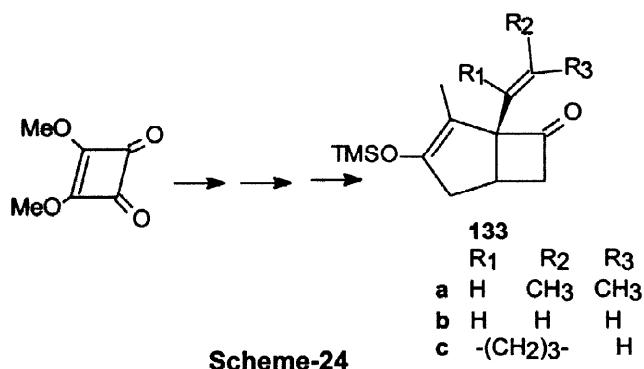


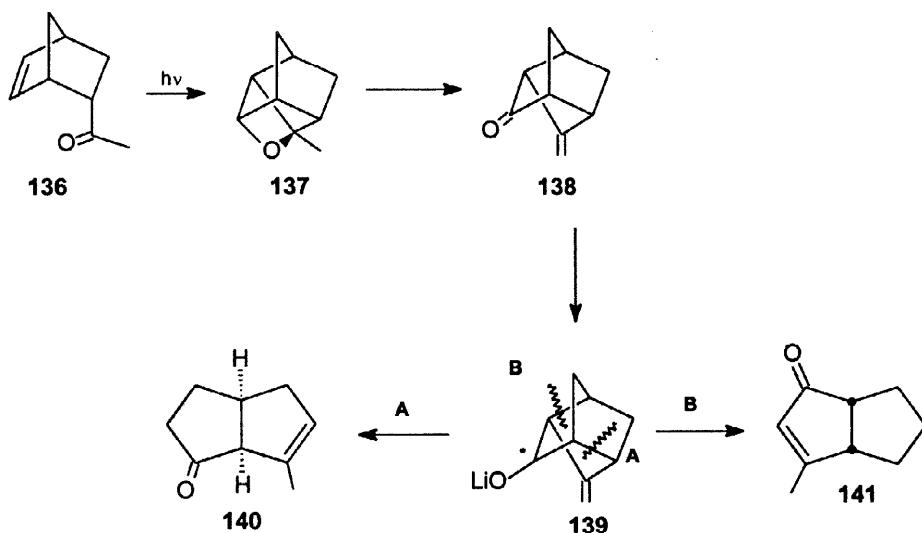
Table-2

2.8. PHOTOCYCLOADDITION-FRAGMENTATION APPROACH

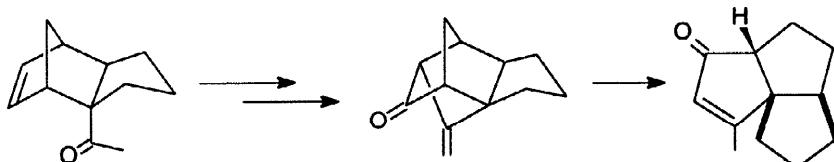
Rawal and coworkers developed an approach involving Diels–Alder reaction to construct first a norbornane skeleton followed by Paterno–Buchi⁶⁴ reaction for the construction of triquinane skeletons **140** and **141**, respectively (Scheme-25). The Paterno–Buchi reaction of the norbornane skeleton of type **136** generates an oxetane **137** which on cleavage and subsequent oxidation gives the ketoalkene **138**. The susceptibility of the ketoalkene to fragmentation due to the inherent strain energy gives the radical **139** which rearranges *via* either path A or B to yield the quinane **140** and **141**, respectively. Later studies showed that the reductive fragmentation of the cage compounds could be controlled by substituents on the norbornane

skeleton. The initial studies were directed towards diquinane and angular triquinane synthesis.⁶⁵ The angular triquinane was obtained *via* the sequence shown in Scheme-26. Subsequently, the linear triquinane *endo*-hirsutene was synthesized using this methodology (Scheme-27).⁶⁶

Towards the synthesis of *endo*-hirsutene, the precursor **143** was generated from fulvene derivative **142** following Sternbach's procedure.^{66b} Photolysis of **143** gave the oxetane derivative **144** which was cleaved with base and the resulting alcohol was oxidized with PDC to give the compound **145**. Base catalyzed fragmentation of **145** directly gave the triquinane **146** which was transformed into *endo*-hirsutene **147** (Scheme-27).



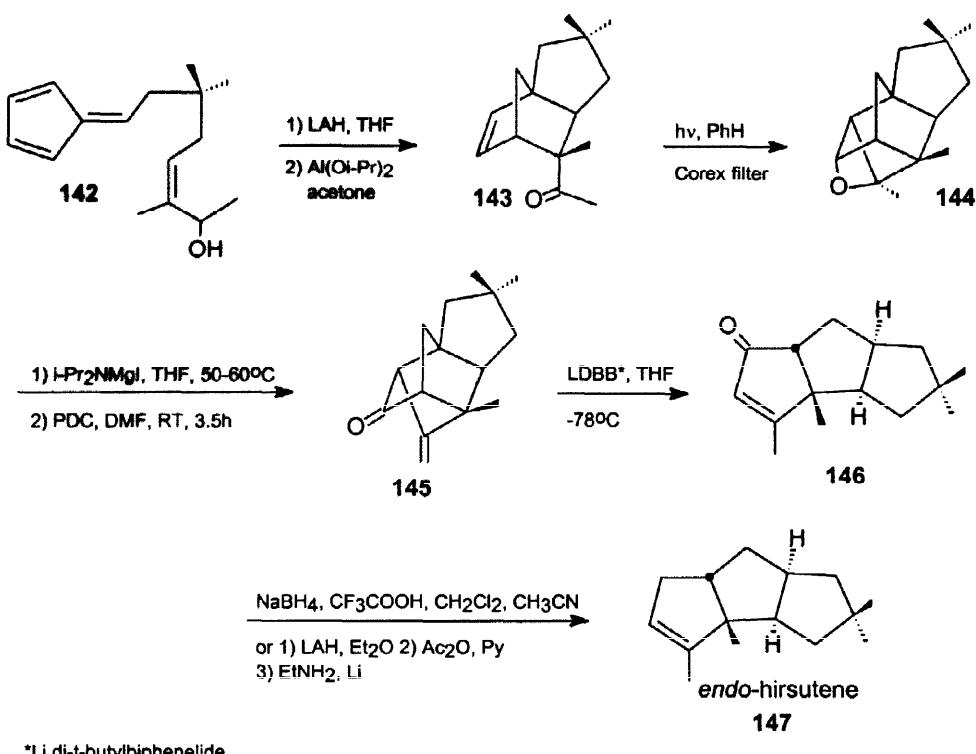
Scheme-25



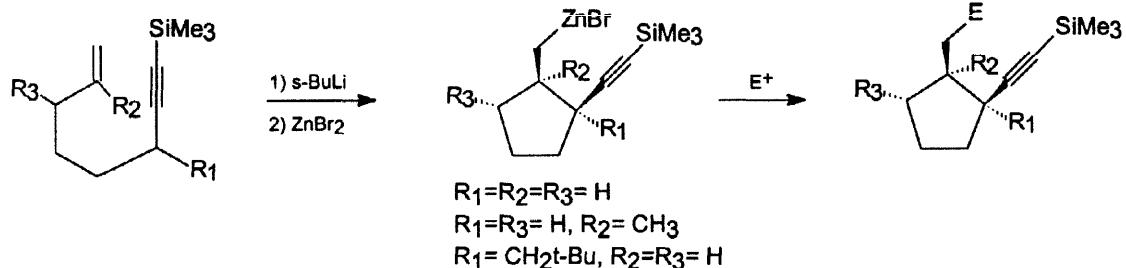
Scheme-26

2.9. ZINCA-ENE-ALLENES REACTION

Meyer and associates⁶⁷ reported a methodology for carbocycle synthesis *via* an intramolecular zinc-ene-allene reaction as shown in Scheme-28. Extension of this methodology provided a route to linearly and angularly fused tricyclopentanoids starting from the same precursor.⁶⁸

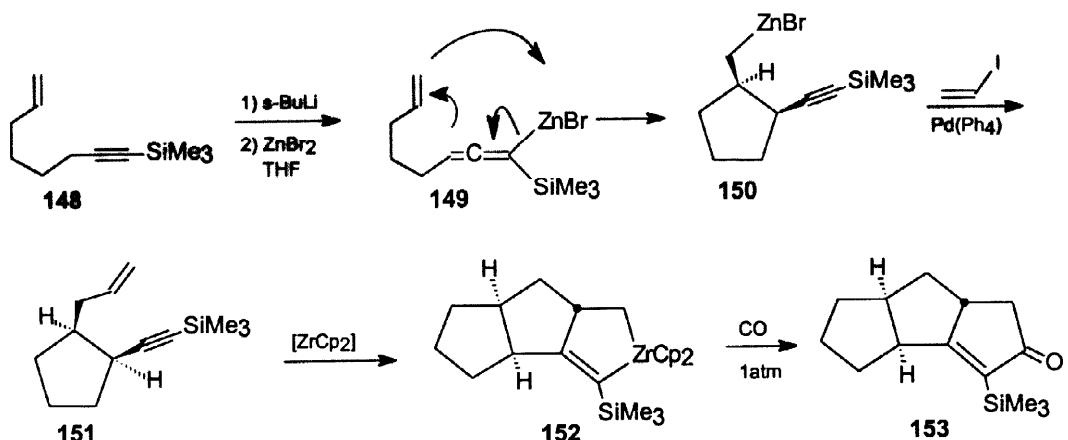


Scheme-27



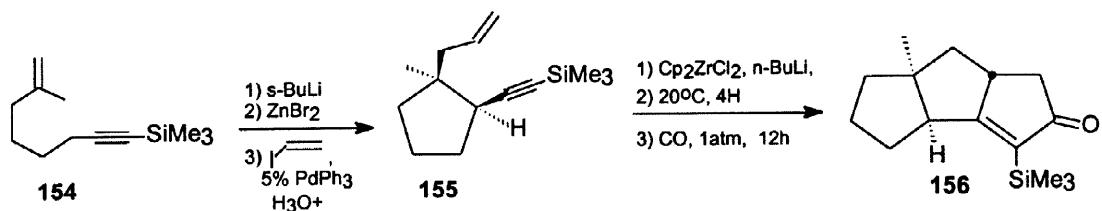
Scheme-28

The authors metallated 8-(trimethylsilyl)-1-octen-7-yne **148** at -45°C with butyllithium and added the Zn salt to the reaction mixture which led to the formation of **150** via the allenic intermediate **149**. The compound **150** was then coupled with 1-iodoethene in presence of tetrakis(triphenylphosphine)palladium to give **151** which on treatment with ZrCp_2 , generated *in situ* from the reaction of ZrCp_2Cl_2 and *n*-butyllithium, gave the zirconatricyclic product **152** which without isolation was subjected to carbonylation reaction to get the *cis:anti:cis* tricyclopentanoid ketone **153**.

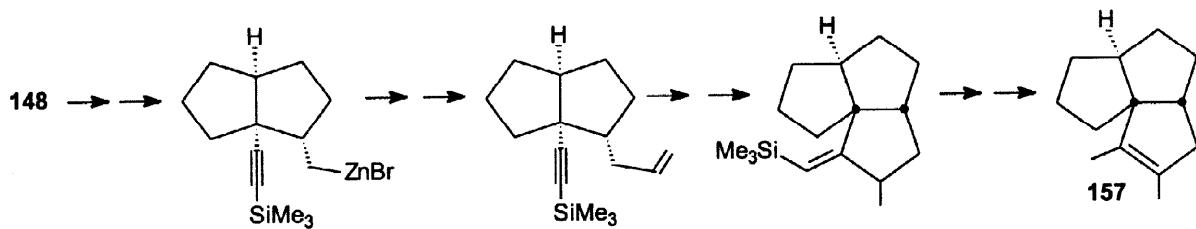


Scheme-29

Utilizing the same methodology α,α -disubstituted enyne **154** was converted to **155** and subsequently to the linear tricyclic enone **156** with an angular methyl group (Scheme-30). Further, this methodology was also used to synthesize an angular triquinane **157** starting from the enyne **148** (Scheme-31).⁶⁸



Scheme-30

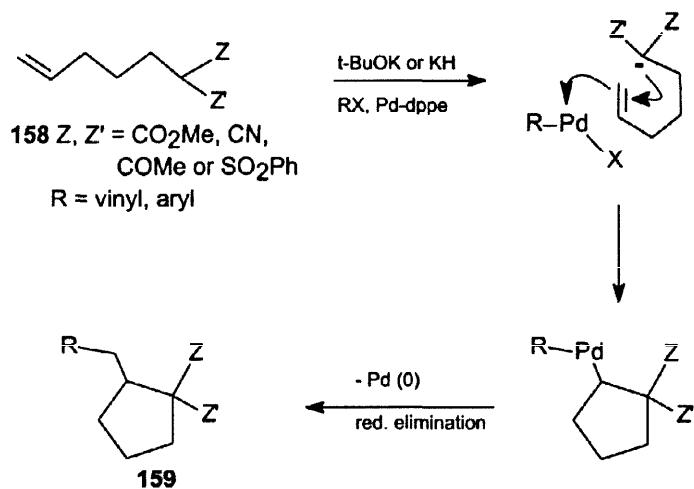


Scheme-31

2.10. PALLADIUM CATALYZED CYCLOPENTANATION OF ALKENES

Balme and his associates have also developed a unified method for rapid synthesis of linear triquinanes employing palladium mediated cyclopentanation of alkenes⁶⁹ bearing a nucleophilic substituent. The methodology is based on their observation that the treatment of *n*-alkene of type **158** with strong base,

palladium catalyst, and an electrophile gives cyclopentane derivatives such as **159** as shown in Scheme-32. Extension of this methodology led to synthesis of capnellene as shown in Scheme-33. The key precursor **160** was prepared from the cyclopentene derivative **160**. Thus, **160** was subjected to iodolactonization and dehydrohalogenation to **161** which was converted to the trisubstituted cyclopentene derivative **163** via **162**. **163** was then subjected to cyclization which gave a mixture of triquinanes **164** and **165** in a 93:7 ratio and good yield. The triquinane **164** was then readily taken to capnellene **8** via **166** as shown in Scheme-33.



Scheme-32

2.11. ONE POT SYNTHESIS *via* CONJUGATE ADDITION OF ENOLATE-CARBANIONS

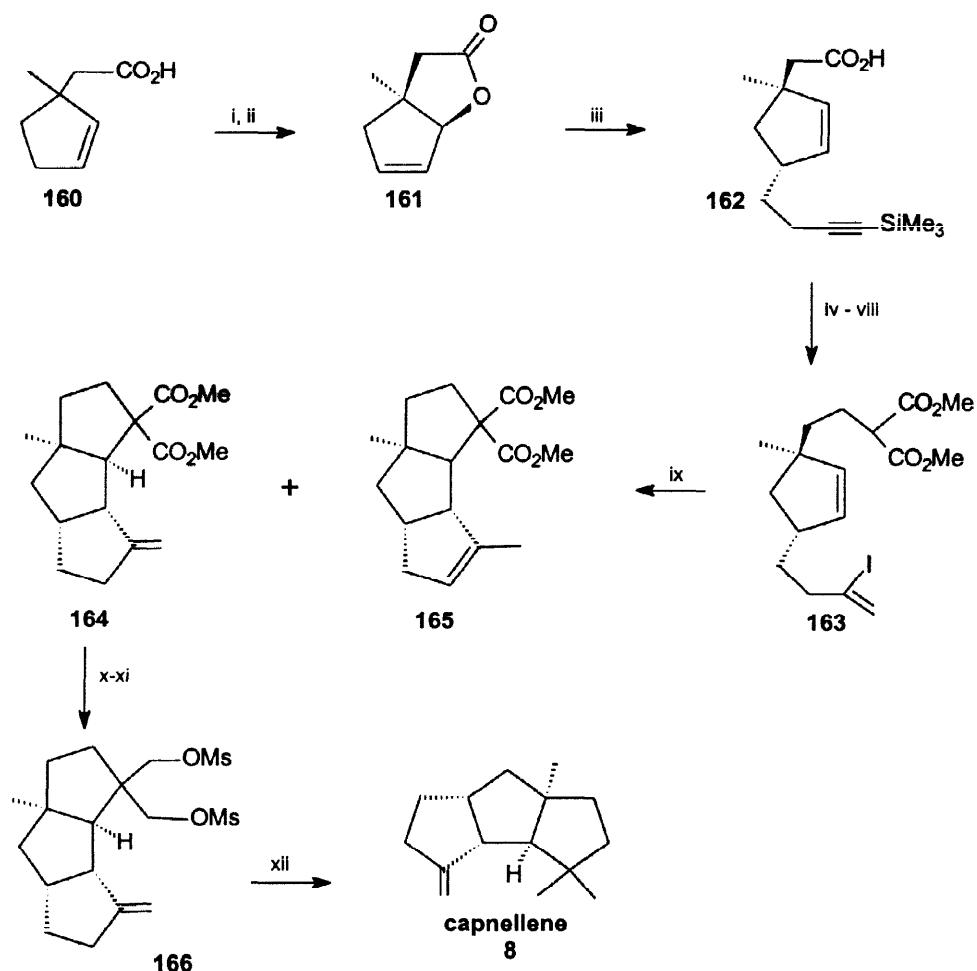
A highly novel and expedient method for a single step, one-pot synthesis of linearly fused triquinanes has been developed by Cohen and coworkers,⁷⁰ which is based on the observation that enolate-carbanions of the type **167** can be generated by the conjugate addition of tris(phenylthio)methyl lithium to an enone⁷¹ and that the lithio thioacetal portion of the resulting species behaves as a nucleophile at -78°C and as an electrophilic carbenoid at slightly higher temperature.⁷² Conjugate addition of **167** to cyclopentenone **168** followed by coupling led to linearly fused *cis:anti:cis* triquinane **169** as shown in the Scheme-34. The mechanism and origin of stereochemistry has been discussed.⁷⁰ The triquinane **169** was transformed into hirsutene as shown in the Scheme-35. Thus, reduction of **169** gave the dione **170** which upon selective ketalization furnished the ketoketal **171**. The carbonyl group was reduced with lithium-ammonia to give the alcohol **172** which was converted into the key intermediate **173**.

2.12 PALLADIUM CATALYZED CYCLIZATION OF ENYNES

Trost and Shi have also developed⁷³ an efficient method for synthesis of a variety of polyquinanes, based on palladium catalyzed cyclization of polyenynes. Thus, when a solution of dienyne **174** was heated in benzene in the presence of $(\text{dba})_3\text{Pd}_2\text{CHCl}_3$, TPP and 10mol% acetic acid, it gave the fully symmetric

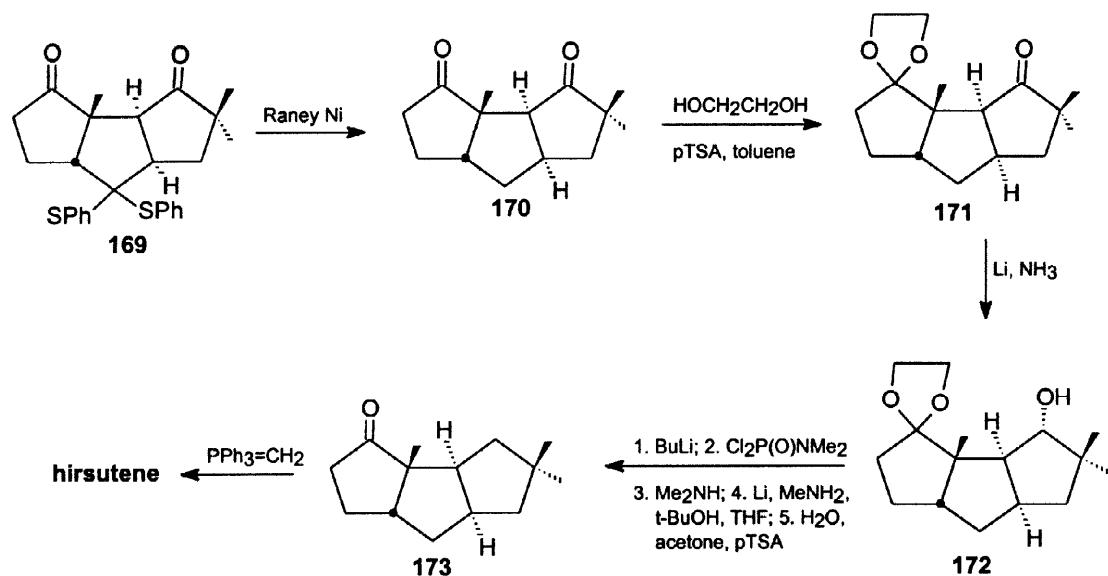
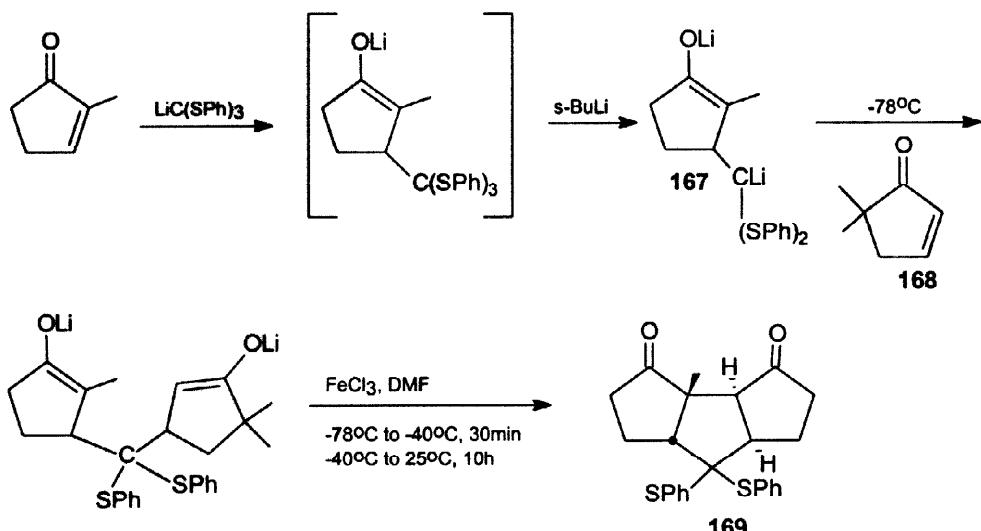
cis:syn:cis triquinane **177** (Scheme-36). The **cis:syn:cis** triquinane **177** is formed via cyclization of **174** to the diquinane-Pd complex **175** which further undergoes Pd initiated cyclization to give the intermediate **176**. The acid catalyzed removal of Pd led to formation of **177**. Similarly, the cyclization of the dienyne **178** in the presence of $(\text{dba})_3\text{Pd}_2\text{CHCl}_3$, PPh_3 in acetic acid gave the propellane type triquinanes **179** which was then converted into **180** (Scheme-37).

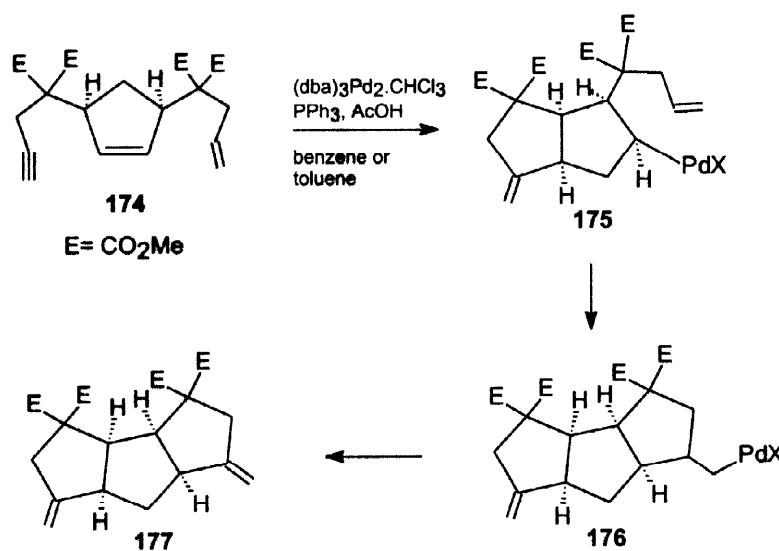
Several other interesting spiro polyquinanes **181-183** (Fig.-3) have also been prepared by cyclization of appropriate enynes. Various aspects of the cyclization mechanistic, regiosomeric and steric have been discussed.⁷³



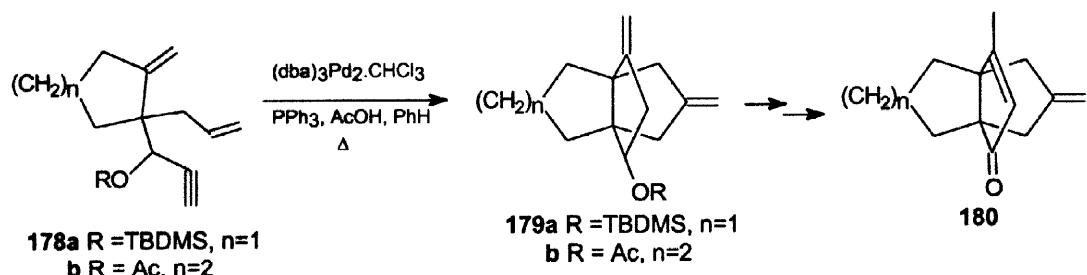
Reagents/Conditions: i) I_2 , KI , NaHCO_3 , H_2O , 20°C ; ii) DBU, toluene, 110°C ; iii) $\text{Me}_3\text{SiC}\equiv\text{CCH}_2\text{CH}_2\text{MgBr}$, $\text{CuBr}\text{-Me}_2\text{S}$, THF, Me_2S , -20°C ; iv) LAH, ether, 0°C ; v) KF , H_2O , DMF; vi) MeSO_2Cl , Et_3N , CH_2Cl_2 , 0°C ; vii) NaH , $\text{CH}_2(\text{CO}_2\text{Me})_2$, 10% Kt , THF, DMF; viii) Me_3SiCl , NaI , H_2O , CH_3CN ; ix) KH , THF, 25°C , 5% $\text{Pd}(\text{OAc})$ 10% tri(2-furyl)phosphine or 10% DPPE x) LAH, Et_2O ; xi) MeSO_2Cl , Et_3N , CH_2Cl_2 ; xii) LiEt_3BH , THF, reflux.

Scheme-33





Scheme-36



Scheme-37

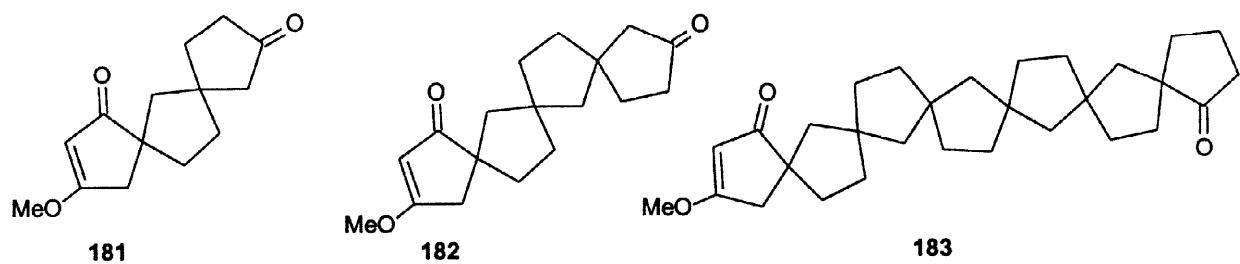
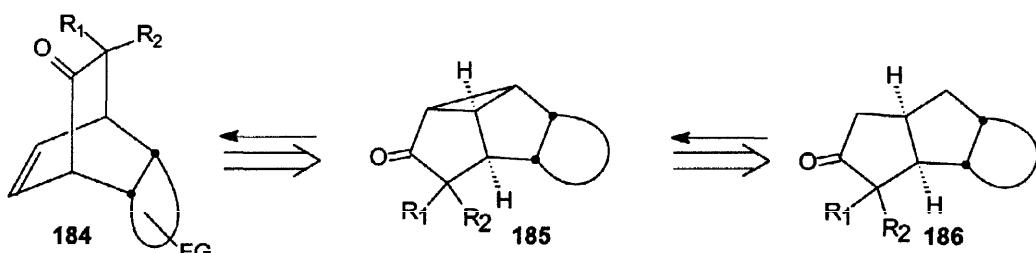


Fig.-3

2.13 DIELS–ALDER CYCLOADDITION-OXA-DI- π -METHANE REARRANGEMENT

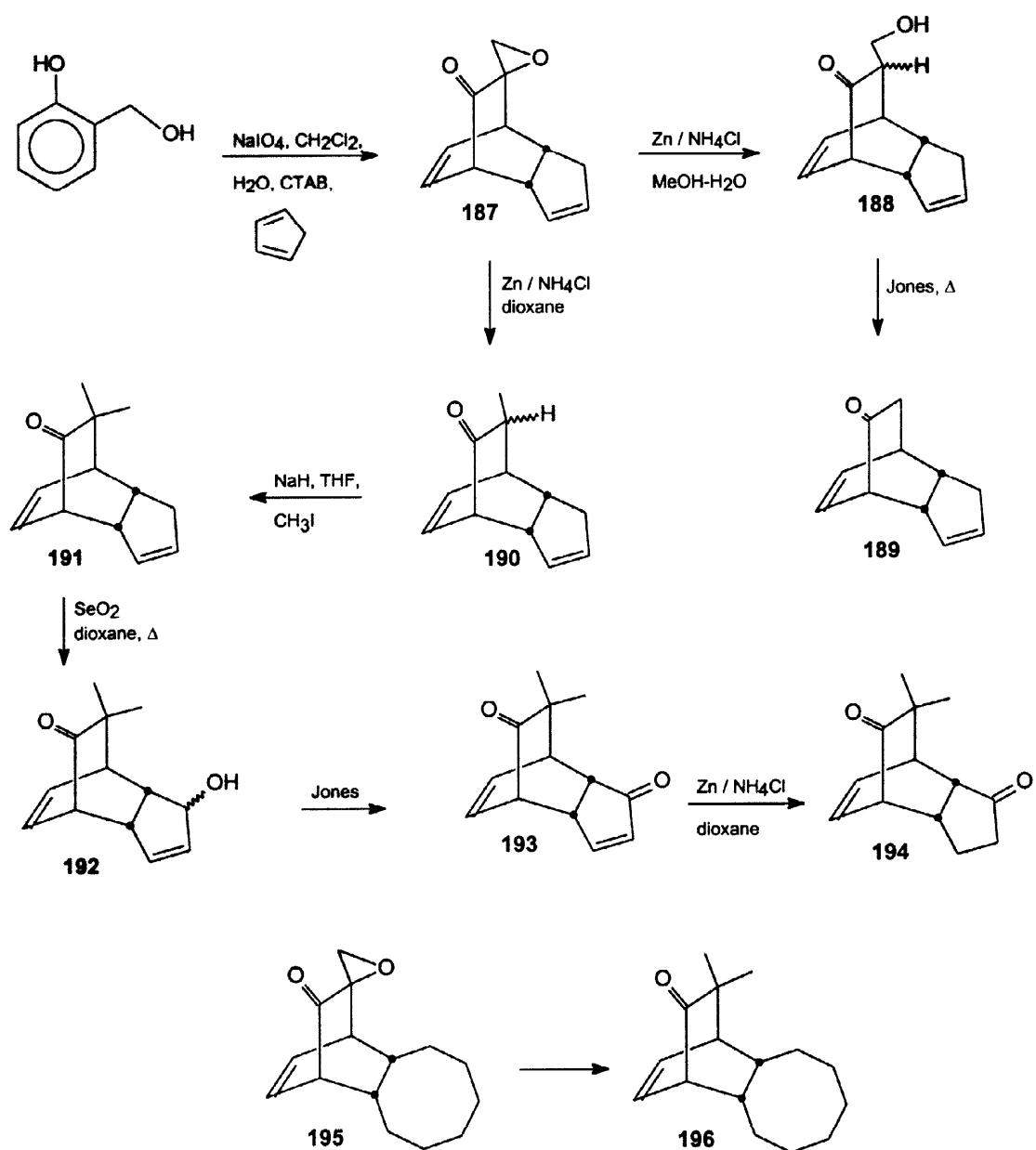
We have also been engaged⁷⁴ in the synthesis of polyquinanes and recently described an efficient and unified strategy for the linearly fused polyquinanes employing inverse demand $\pi^{4s}+\pi^{2s}$ cycloaddition of cyclohexa-2,4-dienones and photochemical oxa-di- π -methane rearrangement as key features of our strategy. Our methodology is based on the idea that linearly fused tricyclopentanoids such as 186 should be readily derived from the tetracyclic system 185, which can be obtained from the *endo*-tricyclic system 184 via oxa-di- π -methane rearrangement (Scheme-38). However, there were no methods for the synthesis of the tricyclic systems of type 184 (Scheme-38). Hence, a general route to annulated bicyclo[2.2.2]octenones from phenols was developed. Thus, several *endo*-annulated tricyclic systems 187-194 were prepared from salicyl alcohol as shown in Scheme-39. The tricyclic system 196 was also obtained from 195 along similar lines. Triplet sensitized photoreaction of 189, 191, 192, 194, 196 in acetone (both sensitizer and solvent) gave the desired tetracyclic systems 197-201 containing the *cis:anti:cis* triquinane framework. The tetracyclic systems 198-201 (Fig.-4) were subjected to reductive cleavage with H₂/Pd-C to give the linearly fused polyquinanes 202-205 in good yield (Scheme-40).^{74g}



Scheme-38

The methodology presented above was successfully applied to an efficient total synthesis⁷⁵ of capnellene, as shown in the Scheme-41. Thus, the *p*-cresol derivative 206 was converted into the annulated bicyclic compound 207 which was transformed into the dimethyl ketone 208. Two fold oxidation of 208 gave the diene-dione 209 which was converted into the keto-ketal 210 following the methodology developed in our group. Triplet sensitized irradiation of the ketal 210 smoothly gave the rearranged product 211. The reductive cleavage of the peripheral cyclopropyl sigma bond gave the triquinane-ketal 212. Barton's deoxygenation of the carbonyl group followed by hydrolysis of the ketal gave the ketone 213 which upon Wittig reaction furnished the natural product capnellene (Scheme-41).

Alternatively, we also discovered a direct route to functionalized *cis:anti:cis* triquinanes of the type 215 via the photoreaction of annulated bicyclo[2.2.2] systems having an α -methoxy- β,γ -enone chromophore such as 214 as shown in the Scheme-42. A number of chromophoric systems have been prepared and rearranged to a variety of functionalized triquinanes 216-218 (Scheme-42).⁷⁶



Scheme-39

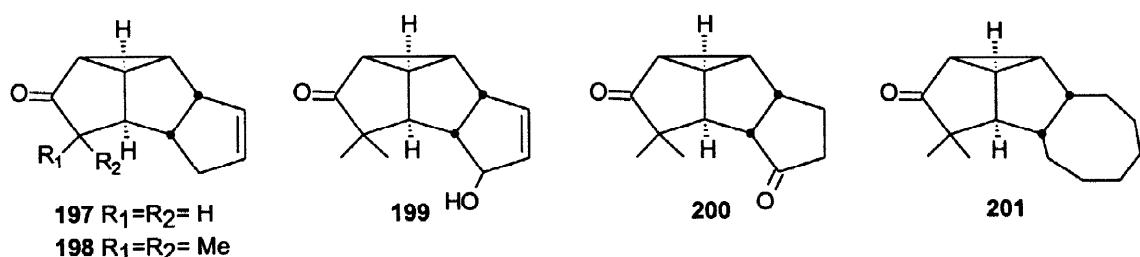
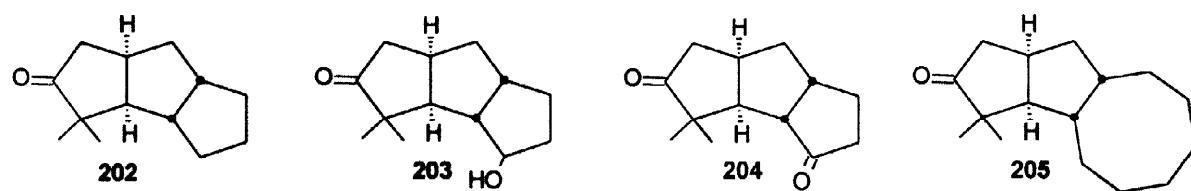
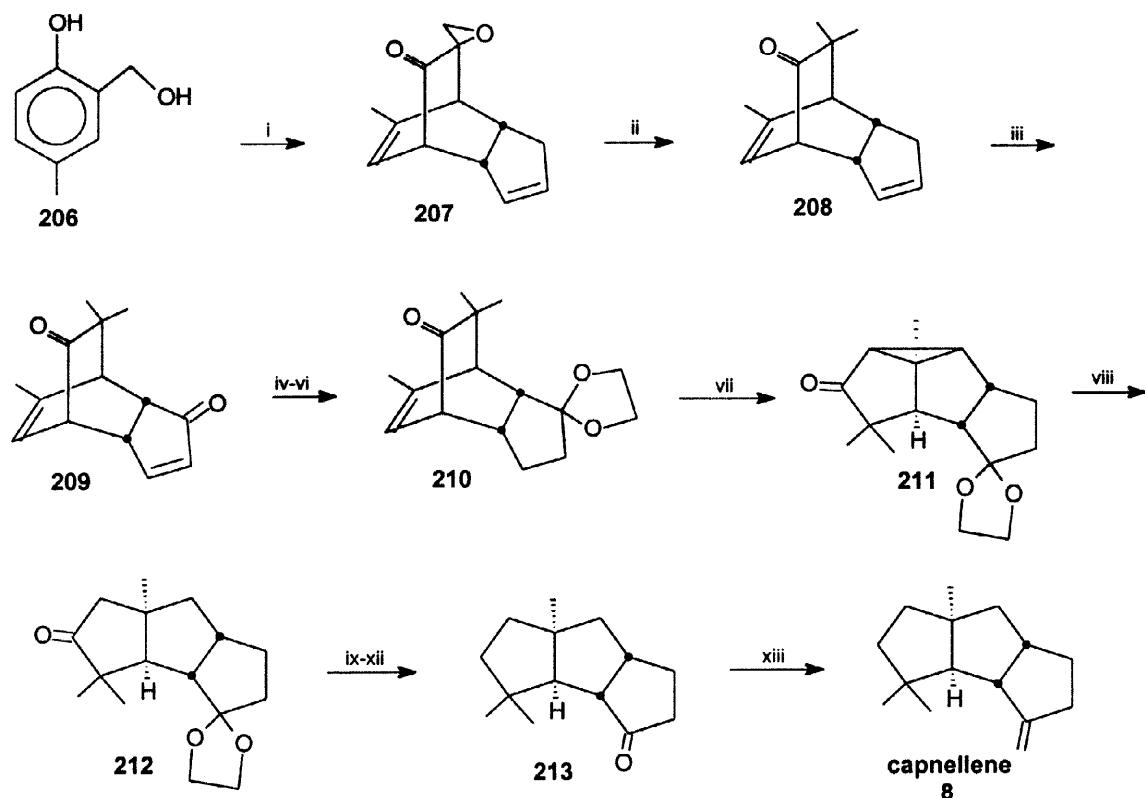


Fig.-4

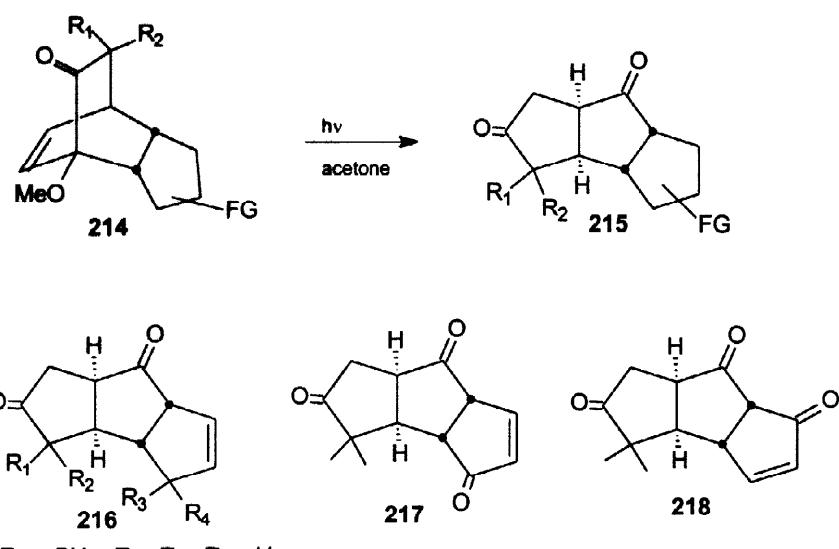


Scheme-40



Reagents/Conditions: i) NaIO_4 , C_5H_6 ; ii) a) Zn , NH_4Cl , dioxane; b) NaH , THF , MeI ; iii) a) SeO_2 , dioxane; b) Jones
iv) NaBH_4 , MeOH ; v) PCC ; vi) ethylene glycol, pTSA ; vii) $\text{h}\nu$, acetone; viii) H_2/Pd ix) NaBH_4 , MeOH ; x) CS_2 ,
 MeI , NaH ; xi) Bu_3SnH ; xii) $\text{H}_2\text{O}/\text{H}^+$; xiii) $\text{CH}_2=\text{PPh}_3$.

Scheme-41



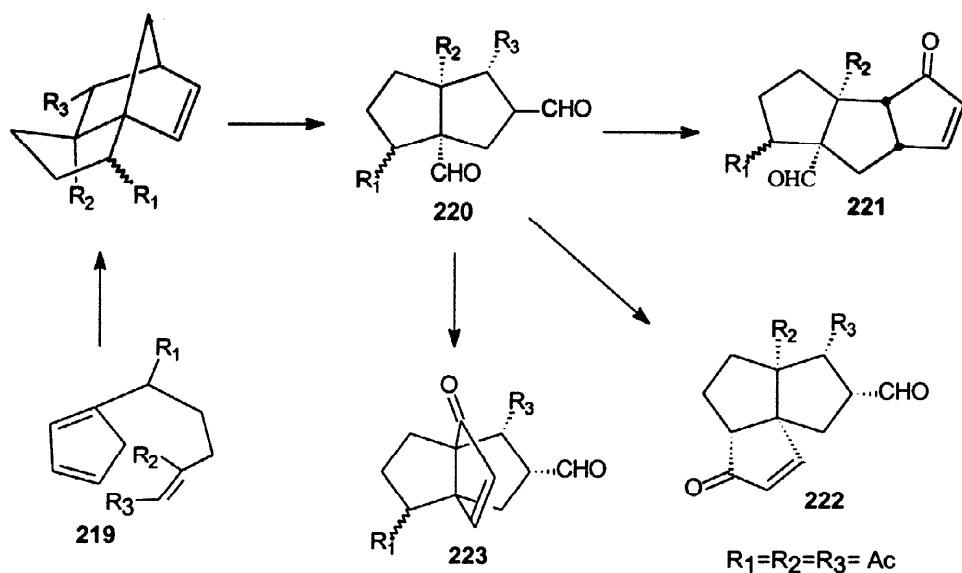
Scheme-4?

3. GENERAL METHODS LEADING TO TRIQUINANES *via* DIQUINANES

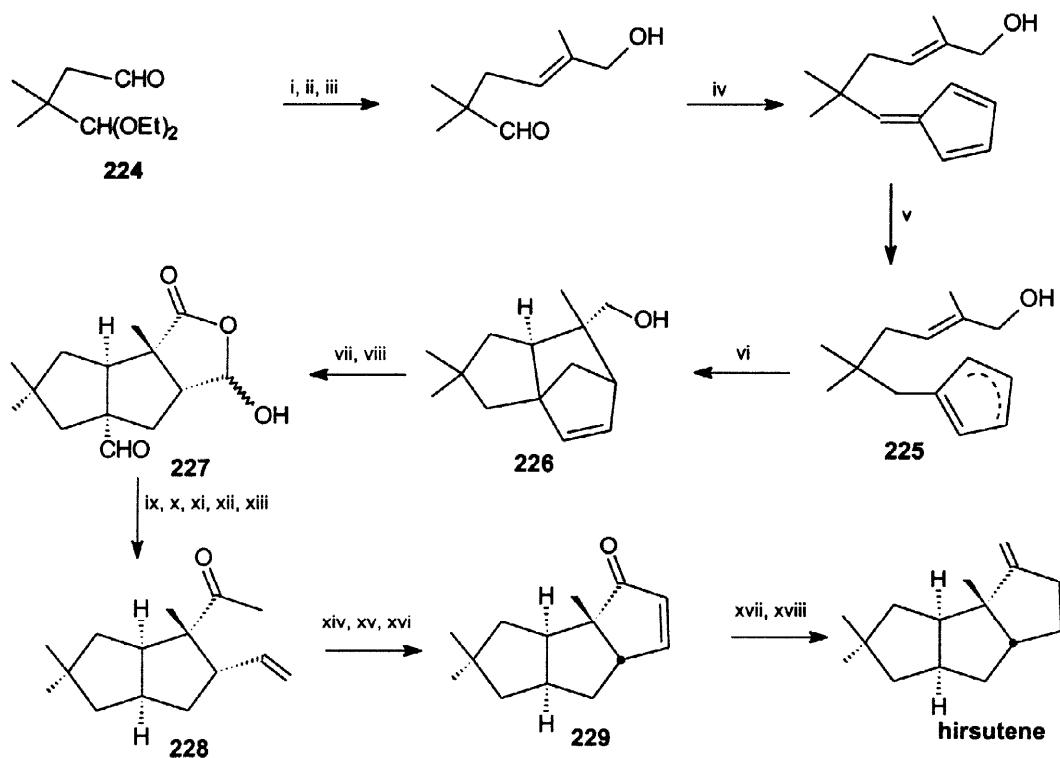
In the previous section we discussed those methods of triquinane syntheses which furnished the linearly fused triquinane frameworks in a single step from appropriate precursors. In this section we present some methods which produce diquinane intermediates which are further annulated to give triquinanes.

3.1 INTRAMOLECULAR DIELS-ALDER APPROACH

Sternbach⁷⁷ and his associates have developed a common strategy for the synthesis of linear (221), angular (222) and propellane type (223) triquinanes employing diquinane intermediates such as 220, readily available through intramolecular Diels–Alder reaction of substituted cyclopentadienes of type 219 as shown in the Scheme-43. In a recent application of this method, Sternbach reported a synthesis of hirsutene^{77a} wherein the desired cyclopentadiene derivative 225 was prepared from the aldehyde 224 as shown in the Scheme-44. The intramolecular Diels–Alder reaction of 225 gave the adduct 226 which was converted into the diquinane 227 by oxidation of the primary alcoholic group to carboxylic acid and cleavage of the double bond. Conversion of the aldehyde 227 to the olefinic ketone 228 and subsequent transformations led to the triquinane 229, which was converted to hirsutene (Scheme-44). The present method thus required 16 steps for the triquinane intermediate (229).



Scheme-43

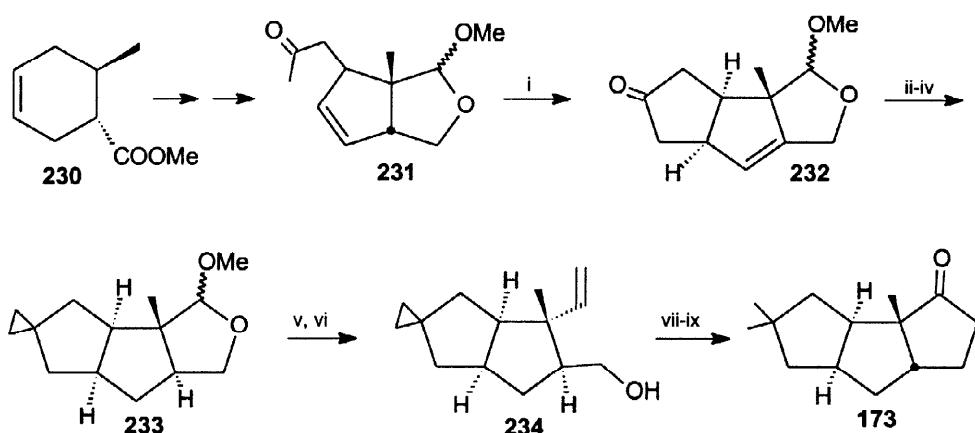


Reagents/Conditions: i) $\text{Ph}_3\text{P}=\text{C}(\text{CH}_3)\text{CO}_2\text{Et}$, PhH, reflux; ii) LAH, ether, 0°C; iii) 10% HCl, THF, r.t.; iv) cyclopentadienylpyrrolidine, Na_2SO_4 , MeOH; v) LAH, THF, r.t.; vi) mesitylene, Δ ; vii) Jones reagent; viii) O_3 , CH_2Cl_2 , -78°C, DMS, -78°C to r.t.; ix) Ac_2O , Et_3N , DMAP; x) Jones reagent; xi) $(\text{COCl})_2$, DMF, benzene, r.t., 2h; xii) 2-mercaptopypyridine-*N*-oxide sodium salt, t-BuSH, DMAP, PhMe, reflux; xiii) MeOH, H_2O , Et_3N (5:4:1) r.t.; xiv) NaH, benzene, r.t., 1h; $(\text{COCl})_2$, r.t., 3h; xvi) Me_2CuLi , ether, -78°C to 0°C; xv) O_3 , CH_2Cl_2 , -78°C, DMS, -78°C; xvii) 5% KOH, ether, THF, n-Bu₄NOH, xviii) PtO₂, H_2 , EtOAc, xviii) $\text{Ph}_3\text{PCH}_2\text{Br}$, KH, PhH, t-BuOH (5:1)

Scheme-44

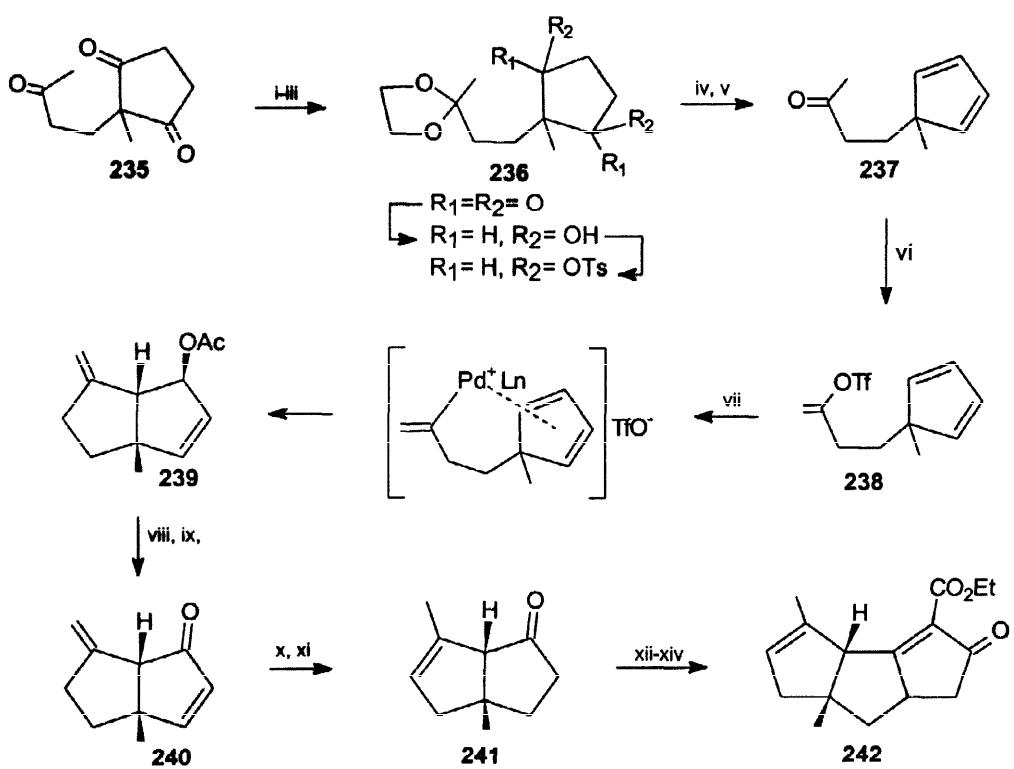
Fukumoto and coworkers⁷⁸ have developed a general route to triquinanes via palladium promoted cyclization to generate diquinane precursors which are further annulated to give linearly fused *cis:anti:cis* tricyclopentanoids and applied their method for total synthesis of hirsutene. The acetal **231** was prepared from cyclohexene derivative **230** in eleven steps and subjected to cyclization in the presence of lithium diisopropylamide, trimethylsilyl chloride and palladium acetate to give the diquinane **232**. The diquinane **232** was converted into the cyclopropyl homologue **233** as shown in the Scheme-45. Elaboration of the hemiacetal **233** to the olefinic alcohol **234** followed by annulation furnished the triquinane intermediate **173** which has been taken to hirsutene.⁷⁰

In context with their continuing interest in polyquinanes,⁷⁹ Shibasaki and coworkers⁸⁰ have reported a synthesis of a triquinane intermediate for capnellenes. Their methodology employed an asymmetric Heck reaction followed by anion capture to generate a diquinane which is annulated to give the tricyclopentanoid. The approach is shown in the Scheme-46. The precursor **238** for the Heck reaction was prepared from the trione **235**. The trione **235** was converted into ditosylate **236**, which upon treatment with DBU and hydrolysis of the ketal group gave the cyclopentadiene derivative **237**. Transformation of **237** to vinyltriflate **238** and Heck reaction in the presence of (S)-BINAP furnished the diquinane **239** in 80% ee. The diquinane **239** was transformed into **240** which after isomerization of the exocyclic double bond led to **241** subsequent annulation of which gave the optically active triquinane **242**, a precursor^{79c,d} for the synthesis of oxygenated capnellenes. Recently synthesis of (–)-capnellene following the above methodology has appeared.^{14g}



Reagents/Conditions: i) LDA, THF, -78°C, TMSCl; Pd(OAc)₂, CH₃CN-CH₂Cl₂; ii) H₂, 10% Pd-C, EtOAc; iii) Ph₃P⁺CH₃Br⁻, n-BuLi, DME, Δ ; iv) CH₂I₂, Et₂Zn, C₆H₆; v) aq. HClO₄, acetone; vi) Ph₃P⁺CH₃Br, n-BuLi, DME, reflux; vii) PCC, NaOAc, CH₂Cl₂; viii) PdCl₂, CuCl, O₂, DMF-H₂O; ix) n-Bu₄N⁺OH⁻, THF-ether, 5% KOH, Δ ; x) H₂, PtO₂, NaOAc, AcOH; xi) PCC, NaOAc, CH₂Cl₂.

Scheme-45



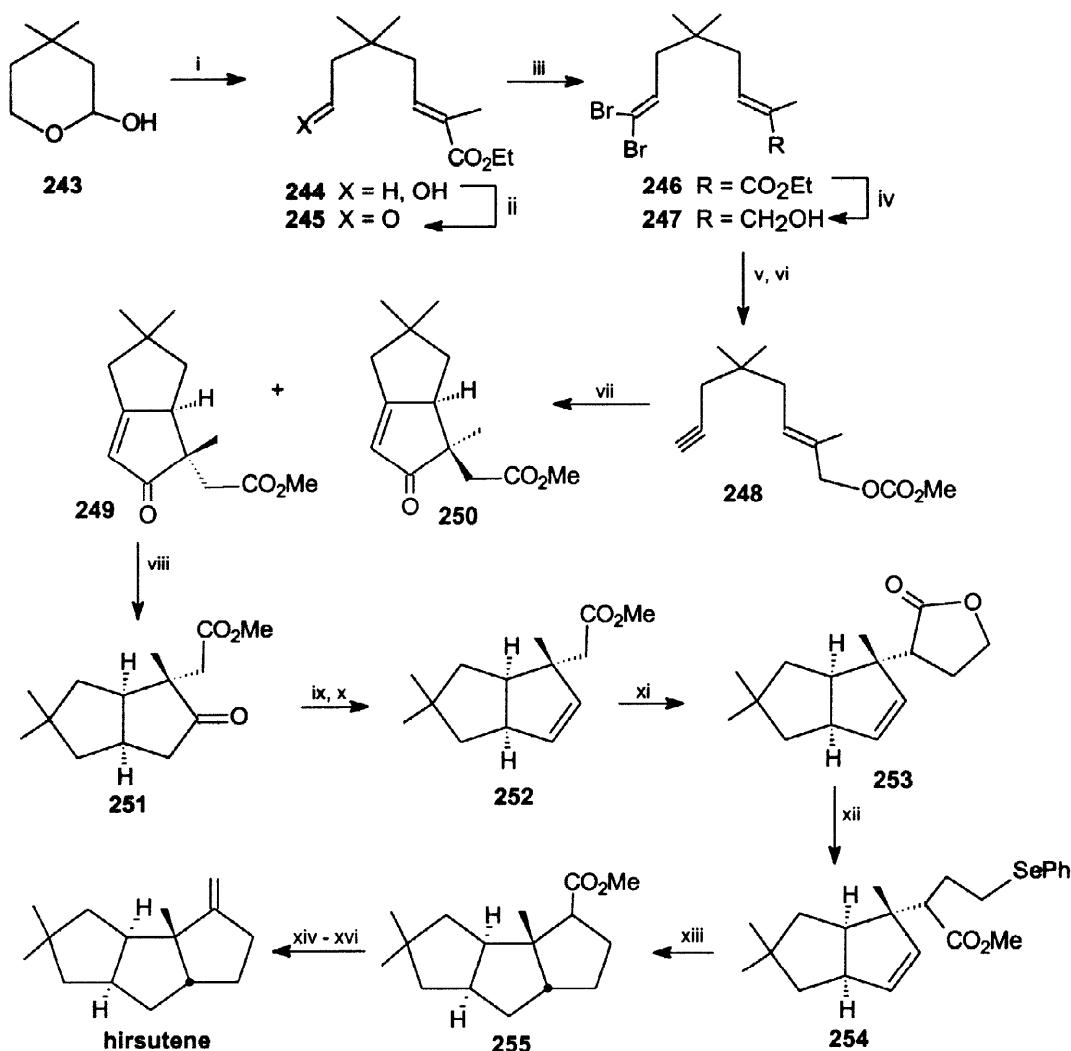
Reagents/Conditions: i) $(\text{TMSCl})_2\text{OTf}$; ii) NaBH_4 ; iii) TsCl , DMAP; iv) DBU; v) TsOH , acetone; vi) LDA, Tf_2NPh ; vii) $\text{Pd}(\text{OAc})_2(\text{S})\text{-BINAP}$, Bu_4NOAc , DMSO; viii) NaOMe ; ix) PDC, MS-3A; x) CuBr , Red-Al, s-BuOH , THF; xi) DBU; xii) LDA, $\text{ICH}_2\text{C}(\text{OMe})=\text{CHCO}_2\text{Et}$; xiii) 30% HClO_4 , ether; xiv) NaOEt , EtOH .

Scheme-46

3.2. PALLADIUM PROMOTED SYNTHESES

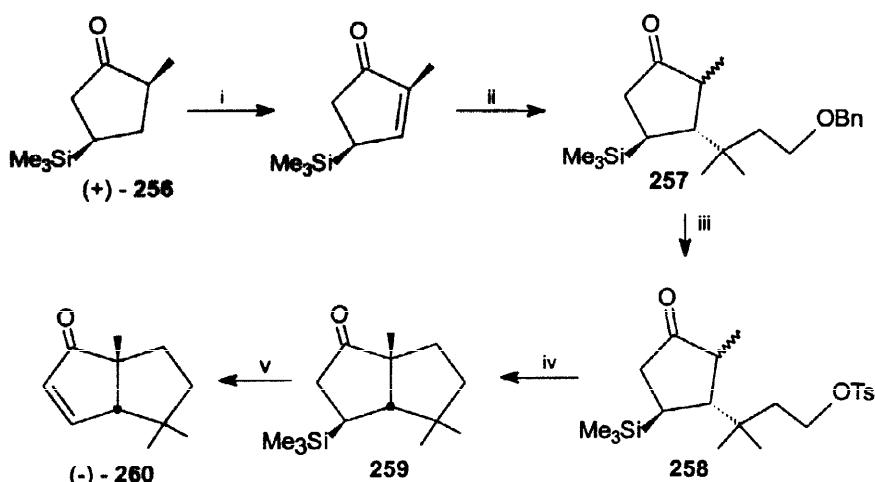
Oppolzer's group developed a method for the synthesis of diquinane intermediates from acyclic enyne systems employing catalytic Pd^0 , Pt^0 and Ni^0 .⁸¹ They recently reported the synthesis of hirsutene^{81c} (Scheme-47) utilizing this allyl-palladium-alkyne cyclization/carbonylation cascade.

The requisite enyne substrate **248** was prepared from the hemiacetal **243** as follows. The hemiacetal **243** was transformed into the alicyclic alcohol **244** and oxidised to **245**. Wittig type condensation of **245** afforded the dihalo-diene ester **246** which was reduced with DIBAH to give the alcohol **247**. Transformation of **247** to the required substrate **248** followed by palladium catalyzed cyclization-carbonylation gave a mixture of diquinanes **249** and **250** (85:15). The major isomer **249** was then hydrogenated to **251** which upon reduction with NaBH_4 and dehydration gave **252**. Conversion of **252** into the lactone **253** and reaction of **253** with sodium phenylselenide followed by esterification with diazomethane gave the selenyl derivative **254** as shown in the Scheme-48. Treatment of **254** with tributyltin hydride gave the tricyclopentanoid **255** bearing all the requisite groups for the synthesis of hirsutene. Reduction of the ester group, transformation to selenocyanate and oxidative elimination of phenylselenyl group furnished the hirsutene.



Scheme-47

Asaoka developed⁸² an enantioselective route to diquinane **260**, thus constituting a formal total synthesis of (–)-capnellene. The route to the diquinane is shown in the Scheme-48. Thus, (+)-2-methyl-4-trimethylsilylcyclopentanone **256** was transformed into the tosylated homologue **258** via benzyl derivative **257**. The intramolecular alkylation of **258** gave the optically active (–)-diquinane **259** which upon oxidative elimination of the silyl group gave **260**. The intermediate **260** has been converted into capnellene by Piers and coworkers.⁸³

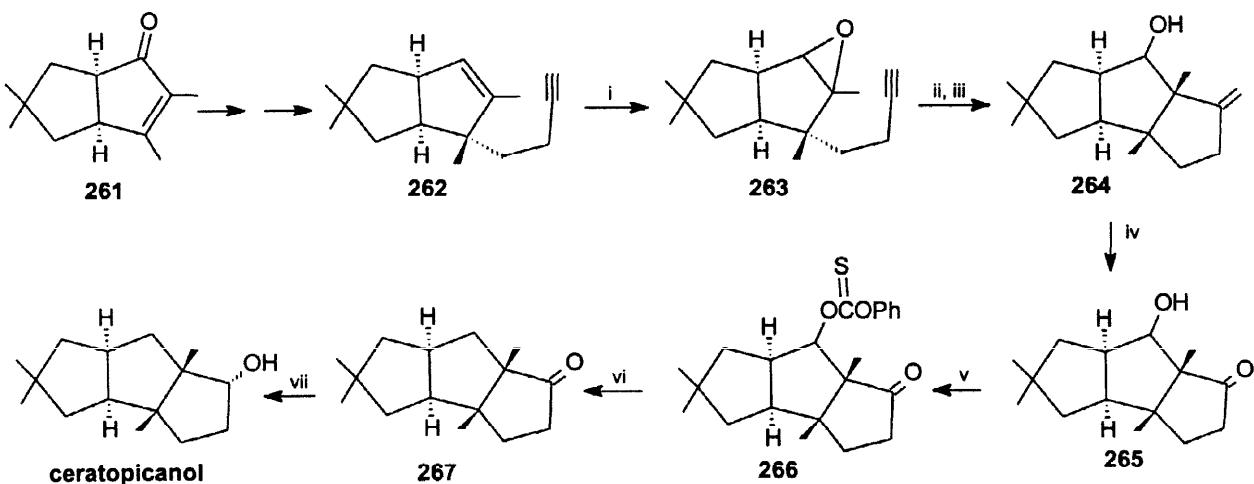


Reagents/Conditions: i) (a) Me_3SiOTf , Et_3N , toluene; (b) $\text{Pd}(\text{OAc})_2$, CH_3CN ,
ii) (a) $\text{BnOCH}_2\text{CH}_2\text{C}(\text{Me})_2\text{MgCl}$, $\text{CuBr} \cdot \text{Me}_2\text{S}$, HMPA, Me_3SiCl , THF; (b) KF , MeOH ;
iii) (a) H_2 , Pd-C , EtOH , (b) TsCl , Py; iv) t-BuOK , THF; v) (a) Me_3SiOTf , Et_3N , toluene,
(b) NBS , THF, (c) TBAF , THF.

Scheme-48

3.3. RADICAL CYCLIZATION

Clive and Magnusson⁸⁴ developed a synthesis of ceratopicanol from a diquinane precursor employing radical cyclization of epoxyalkyne moiety to create a triquinane system (Scheme-49). The precursor **262** was obtained from the diquinane **261** in six steps. Epoxidation of **262** gave the epoxyalkyne **263** which was treated with Cp_2TiCl to give the triquinane alcohol **264** after an acidic workup. Protection of the hydroxyl group, cleavage of the exocyclic double bond and deprotection gave the keto alcohol **265**. Deoxygenation of the hydroxyl group via thiocarbonate **266** furnished the ketone **267**, a precursor which was taken to ceratopicanol.



Reagents/Conditions: i) mcpba, CH_2Cl_2 , 0°C , 2h; ii) Cp_2TiCl , THF, rt, 7h; iii) 10% H_2SO_4 workup; iv) Ac_2O , AcCl , DMAP, Py, 25°C , 2-4h; OsO_4 , 4-methylmorpholine-*N*-oxide, 10:1 acetone- H_2O , 12h; K_2CO_3 , MeOH , 1h; $\text{Pb}(\text{OAc})_4$, K_2CO_3 , CH_2Cl_2 , 0°C to 25°C ; v) $\text{ClC}(\text{S})\text{OPh}$, DMAP, CH_3CN , 8h; vi) Bu_3SnH , Et_3B , air, hexane, 0°C , 1h, workup and treatment with stannane/borane system; vii) NaBH_4 , MeOH , -20°C , 20min.

Scheme-49

4. SUMMARY AND OUTLOOK

This review highlights that the realm of polyquinane chemistry continues to hold synthetic challenge to the organic chemists. In the previous decade most of the syntheses were multistep sequences directed towards linearly fused cyclopentanoids. The focus now appears to have shifted towards development of general protocols to achieve the syntheses of a large number of triquinanes *via* a short and stereoselective sequence. The goal being to synthesize the *cis:anti:cis* linear triquinane framework in a single step from a precursor embodying, as far as possible, all the requisite carbon atoms and functionalities such that the introduction of functional groups with the correct stereochemistry in the precursor determines the final outcome in the product. The methodologies described in this review are among the efforts made in this direction.

ACKNOWLEDGMENT: We thank the Department of Science and Technology for financial support. One of us (B. T.) is thankful to CSIR, New Delhi for a fellowship.

REFERENCES

1. (a) Devon, T.K., Scott, A.I. *Handbook of Naturally Occurring Compounds*; Academic Press: New York, 1972; Vol. II, 180. (b) Nakanishi, K., Goto, T.; Ito, S.; Natori, S.; Nozoe, S. *Natural Products Chemistry*; Kodansha Ltd.: Tokyo; 1983, Vol. III, Chapter 2, 51-56. (c) Comer, F. W.; Trotter, J. *J. Chem. Soc. B*, 1966, 11-18.
2. (a) Bohlman, F.; Jakupovic, J. *Phytochemistry* 1980, 19, 259-265. (b) Dominguez, K. A.; Cano, C.; Franco, R.; Villareal, A. M.; Watson, W. H.; Zabel, V. *Phytochemistry* 1980, 19, 2478. (c) Corbett, R. E.; Lauren, D. R.; Weavers, R. T. *J. Chem. Soc., Perkin Trans. 1* 1979, 1774-1790. (d) Kaneda, M.; Takahashi, R.; Itaka, Y.; Shibata, S. *Tetrahedron Lett.* 1972, 13, 4609-4611.
3. (a) Sheikh, Y. M.; Singy, G.; Kaisin, M.; Eggert, H.; Djerassi, C.; Tursch, B.; Daloze, D.; Braekman, J. C. *Tetrahedron* 1976, 32, 1171-1178. (b) Kaisin, M.; Sheikh, Y. M.; Durham, L. J.; Djerassi, C.; Tursch, B.; Daloze, D.; Braekman, J. C.; Losman, D.; Karlsson, R. *Tetrahedron Lett.* 1974, 15, 2239-2242. (c) Ayanoglu, E.; Gebreyesus, T.; Beechan, C. M.; Djerassi, C.; Kaisin, M. *Tetrahedron Lett.* 1978, 19, 1671-1674.
4. (a) Martin, D. G.; Slomp, G.; Miszak, S.; Duchamp, D. J.; Chidester, C. G. *Tetrahedron Lett.* 1970, 4901-4904. (b) Comer, F. W.; McCapra, F.; Qureshi, I. H.; Scott, A. I. *Tetrahedron* 1967, 23, 4761-4768. (c) Shuji, T.; Naganawa, H.; Iinuma, H.; Takita, T.; Maeda, K.; Umezawa, H. *Tetrahedron Lett.* 1971, 1955-1958. (d) Nakamura, H.; Takita, T.; Umezawa, H.; Kunishima, M.; Nakayama, Y.; Itaka, Y. *J. Antibiot.* 1974, 27, 301-302. (e) Nozoe, S.; Furukawa, J.; Sankawa, U.; Shibata, S. *Tetrahedron Lett.* 1976, 17, 195-198. (f) Ranieri, R. L.; Calton, G. J.; *Tetrahedron Lett.*, 1978, 19, 499-502.
5. Paquette, L. A. *Topics in Current Chemistry*; Springer-Verlag, 1984, 119, 1-146.

6. Paquette, L. A. *Topics in Current Chemistry*; Springer-Verlag, 1979, 79, 41-165.
7. Osawa, E.; Aigami, K.; Takaishi, N.; Inamoto, Y.; Fujikura, Y.; Majerski, Z.; Schleyer, P. v. R.; Engler, E. M.; Farcasiu, M. *J. Am. Chem. Soc.* 1977, 99, 5361-5373.
8. (a) Eaton, P. E.; Mueller, R. H.; Carlson, G. R.; Cullison, D. A.; Cooper, G. F.; Chou, T. C.; Krebs, E.-P. *J. Am. Chem. Soc.* 1977, 99, 2751-2767. (b) Eaton, P. E.; Mueller, R. H. *J. Am. Chem. Soc.* 1972, 94, 1014-1016. (c) Paquette, L. A.; Balogh, D. W.; Usha, R.; Kountz, D.; Christopher, G. C. *Science* 1981, 211, 575-576.
9. Mehta, G.; Prabhakar, C. *J. Org. Chem.* 1995, 60, 4638-4640.
10. (a) Burkholder, P. R.; Burkholder, L. M. *Science* 1958, 127, 1174-1175. (b) Tursch, B.; Braekman, J. C.; Dalzoe, D., Kaisin, M. *Marine Natural Products*; Scheuer, P. J.; Ed.; Academic Press: New York, 1978, Vol. 2, 261-262.
11. (a) Paquette, L. A.; Doherty, A. M. *Polyquinane Chemistry*; Hafner, L.; Rees, C. W.; Trost, B. M.; Lehn, J. M. ; Schleyer, P. v. R.; Zahradnik, R.; Eds.; Springer-Verlag: New York, 1987, Vol. 26, 1-225. (b) Hudlicky, T.; Rulin, F.; Lovelace, T. C.; Reed, J. W. *Studies in Natural Products Chemistry*; Rahman, A.; Ed.; Elsevier: Amsterdam, 1989, Vol. 3, 3-72. (c) After submission of this manuscript a comprehensive account on the synthesis of polyquinane natural products appeared : Mehta, G.; Srikrishna, A. *Chem. Rev.* 1997, 97, 671-719.
12. (a) Vandewalle, M.; Clercq, P. D. *Tetrahedron* 1985, 41, 1767-1831. (b) Trost, B. M. *Chem. Soc. Rev.* 1982, 11, 141-170. (c) Graham, S. L.; Heathcock, C. H.; Pirrung, M. C.; Plavac, F.; White, C. T. *The Total Synthesis of Natural Products*; ApSimon, J.; Ed.; Wiley: New York, 1983; Vol.5, 318-322. (c) Mehta, G. *J. Sci. Ind. Res. (India)* 1978, 37, 256-264.
13. Hudlicky, T.; Price, J. D. *Chem. Rev.* 1989, 89, 1467-1486.
14. (a) Jung, M. E.; Rayle, H. L. *J. Org. Chem.* 1997, 62, 4601-4609. (b) Tormo, J.; Moyano, A.; Pericas, M. A.; Riera, A. *J. Org. Chem.* 1997, 62, 4851-4856. (c) Baralotto, C.; Chanon, M.; Julliard, M. *J. Org. Chem.* 1996, 61, 3576-3577. (d) Coates, R. M.; Ho, Z.; Zhu, L. *J. Org. Chem.* 1996, 61, 1184-1186. (e) Doi, T.; Yanagisawa, A.; Nakanishi, S.; Yamamoto, K.; Takahashi, T. *J. Org. Chem.* 1996, 61, 2602-2603. (f) Lee, H-Y; Kim, D. I. *J. Chem. Soc., Chem. Commun.* 1996, 1539-1540. (g) Ohshima, T.; Kagechika, K.; Adachi, M.; Sodeoka, M.; Shibasaki, M. *J. Am. Chem. Soc.* 1996, 118, 7108-7116.
15. Inoue, T.; Hosomi, K.; Araki, M.; Nishide, K.; Node, M. *Tetrahedron Assym.* 1995, 6, 31-34.
16. Takemoto, Y.; Furuse, S.; Koike, H.; Ohra, T.; Iwata, C.; Ohishi, H. *Tetrahedron Lett.* 1995, 36, 4085-4088.
17. (a) Little, R. D.; Bukhari, A.; Venegas, M. G. *Tetrahedron Lett.* 1979, 20, 305-308. (b) Venegas, M. G.; Little, R. D. *Tetrahedron Lett.* 1979, 20, 309-312. (c) Little, R. D.; Carroll, G. L. *J. Org. Chem.* 1979, 44, 4720-4722.

18. (a) Little, R. D.; Muller, G. W. *J. Am. Chem. Soc.* **1979**, *101*, 7129-7130. (b) Little, R. D.; Muller, G. W.; Venegas, M. G.; Carroll, G. L.; Bukhari, A.; Patton, L.; Stone, K. *Tetrahedron* **1981**, *37*, 4371-4383. (c) Little, R. D.; Higby, R. G.; Moeller, K. D. *J. Org. Chem.* **1983**, *48*, 3139-3140.
19. (a) Hijfte, L. V.; Little, R. D. *J. Org. Chem.* **1985**, *50*, 3940-3942. (b) Hijfte, L. V.; Little, R. D.; Petersen, J. L.; Moeller, K. D. *J. Org. Chem.* **1987**, *52*, 4647-4660. (c) Kissel, C. L.; Rickborn, B. J. *Org. Chem.* **1972**, *37*, 2060-2063. (d) Crandall, J. K.; Lin, L. C. *J. Org. Chem.* **1968**, *33*, 2375-2378.
20. (a) Little, R. D.; Carroll, G. L. *Tetrahedron Lett.* **1981**, *22*, 4389-4392. (b) Little, R. D.; Carroll, G. L.; Petersen, J. L. *J. Am. Chem. Soc.* **1983**, *105*, 928-932. (c) Little, R. D.; Masjedizadeh, M. R.; Moeller, K. D.; Dannecker-Doerig, I. *Synlett* **1992**, 107-113. (d) Little, R. D. *Chem. Rev.* **1996**, *96*, 93-114.
21. Berson, J. A. *Acc. Chem. Res.* **1978**, *11*, 446-453.
22. (a) Baseman, R. J.; Pratt, D. W.; Chow, M.; Dowd, P. *J. Am. Chem. Soc.* **1976**, *98*, 5726-5727. (b) Trost, B. M.; Chan, D. M. T. *J. Am. Chem. Soc.* **1980**, *102*, 6359-6361.
23. Koreeda, M.; Mislanker, S. G. *J. Am. Chem. Soc.* **1983**, *105*, 7203-7205.
24. (a) Bryce-Smith, D.; Gilbert, A.; Orger, B. H. *Chem. Commun.* **1966**, 512-513. (b) Wilzbach, K. E.; Kaplan, L. *J. Am. Chem. Soc.* **1966**, *88*, 2066-2067. (c) Bryce-Smith, D.; Foulger, P.; Forrester, J.; Gilbert, A.; Orger, B. H.; Tyrrell, H. M. *J. Chem. Soc., Perkin Trans. 1* **1980**, 55-71. (d) Gilbert, A.; Yianni, P. *Tetrahedron* **1981**, *37*, 3275-3283.
25. Ferree, W. J.; Grutzner, J. B.; Morrison, H. *J. Am. Chem. Soc.* **1971**, *93*, 5502-5512.
26. Wender, P. A.; Siggel, L.; Nuss, J. M. *Comprehensive Organic Synthesis*; Trost, B. M.; Fleming, I. A.; Eds.; Academic Press: New York, **1991**; Vol. 5, 645-673.
27. Bryce-Smith, D.; Gilbert, A.; Mattay, J. *Tetrahedron* **1986**, *42*, 6011-6014.
28. (a) Mattay, J. *Tetrahedron* **1985**, *41*, 2405-2417. (b) Mattay, J. *Tetrahedron* **1985**, *41*, 2393-2404.
29. Mattay, J.; Russink, J.; Piccirilli, J. A.; Jans, A. W. H.; Cornelisse, J. *J. Chem. Soc., Perkin Trans. 1* **1987**, 15-20.
30. Wender, P. A.; Howbert, J. J. *J. Am. Chem. Soc.* **1981**, *103*, 688-690.
31. Wender, P. A.; Dreyer, G. B. *Tetrahedron* **1981**, *37*, 4445-4450.
32. Wender, P. A.; Ternansky, R. J. *Tetrahedron Lett.* **1985**, *26*, 2625-2628.
33. Wender, P. A.; Singh, S. K. *Tetrahedron Lett.* **1985**, *26*, 5987-5990.
34. (a) Wender, P. A.; Howbert, J. J. *Tetrahedron Lett.* **1982**, *23*, 3983-3986. (b) Danishefsky, S. Zamboni, R.; Kahn, M.; Etheredge, S. J. *J. Am. Chem. Soc.* **1980**, *102*, 2097-2098.
35. (a) Mehta, G.; Murthy, A. N.; Reddy, D. S. K.; Reddy, A. V. *J. Am. Chem. Soc.* **1986**, *108*, 3443-3452. (b) Mehta, G.; Srikrishna, A.; Reddy, A. V.; Nair, M. S. *Tetrahedron* **1981**, *37*, 4543-4559.
36. Mehta, G.; Reddy, D. S.; Murthy, A. N. *J. Chem. Soc., Chem. Commun.* **1983**, 824-825.

37. (a) Stevens, K. E.; Paquette, L. A. *Tetrahedron Lett.* **1981**, *22*, 4393-4396. (b) Stevens, K. E.; Paquette, L. A. *Can. J. Chem.* **1984**, *62*, 2415-2420. (c) Magnusson, G.; Thoren, S. *J. Org. Chem.* **1973**, *38*, 1380-1384.
38. Mehta, G.; Reddy, A. V.; Murthy, A. N.; Reddy, D. S. *J. Chem. Soc., Chem. Commun.* **1982**, 540-541.
39. (a) Iseki, K.; Yamazaki, M.; Shibasaki, M.; Ikegami, S. *Tetrahedron* **1981**, *37*, 4411-4418. (b) Shibasaki, M.; Iseki, K.; Ikegami, S. *Tetrahedron Lett.* **1980**, *21*, 3587-3590.
40. Mehta, G.; Reddy, A. V. *J. Chem. Soc., Chem. Commun.* **1981**, 756-757.
41. (a) Curran, D. P.; Rekiewicz, D. M. *J. Am. Chem. Soc.* **1985**, *107*, 1448-1449. (b) Ryu, I.; Sonoda, N.; Curran, D. P. *Chem. Rev.* **1996**, *96*, 177-194.
42. Curran, D. P.; Rakiewicz, D. M. *Tetrahedron* **1985**, *41*, 3943-3948.
43. Beckwith, A. L. J.; Ingold, K. U. *Rearrangements in Ground and Excited States*; de Mayo, P.; Ed; Academic Press: New York, **1980**, Vol.1, 161-283.
44. (a) Beckwith, A. L. J. *Tetrahedron* **1981**, *37*, 3073-3100. (b) Surzur, J. M. *Reactive Intermediates*, Abramovitch, R. A.; Ed; Plenum Press: New York, **1982**, Vol.2, Chapter 3, 121-277, .
45. Beckwith, A. L. J.; Schiesser, C. H. *Tetrahedron* **1985**, *41*, 3925-3941.
46. Curran, D. P.; Chen, M. H. *Tetrahedron Lett.* **1985**, *26*, 4991-4994.
47. Fevig, T. L.; Elliot, R. L.; Curran, D. P. *J. Am. Chem. Soc.* **1988**, *110*, 5064-5067.
48. Khand, I. U.; Knox, G. R.; Pauson, P. L.; Watts, W. E.; Foreman; M. I. *J. Chem. Soc., Perkin Trans. 1* **1973**, 977-981.
49. (a) Pauson, P. L. *Tetrahedron* **1985**, *41*, 5855-5860. (b) Schore, N. E. *Comprehensive Organic Synthesis*; Trost, B. M.; Fleming, I. A.; Eds.; Academic Press: New York, **1991**; Vol. 5, 1037-1064.
50. Schore, N. E. *Organic Reactions* **1990**, *40*, 1-90.
51. Schore, N. E.; Croudace, M. C. *J. Org. Chem.* **1981**, *46*, 5436-5438.
52. Exon, C.; Magnus, P. *J. Am. Chem. Soc.* **1983**, *105*, 2477-2478.
53. Magnus, P.; Exon, C.; Albaugh-Robertson, P. *Tetrahedron* **1985**, *41*, 5861-5869.
54. Gybin, A. S.; Smit, W. A.; Caple, R.; Veretenov, A. S.; Shaskov, A.; Vorontsova, L. G.; Kurella, M. G.; Chertkov, V. S.; Carapetyan, A. A.; Kosnikov, A. Y.; Alexanyan, M. S.; Lindeman, S. V.; Panov, V. N.; Maleev, A. V.; Struchkov, Y. T.; Sharpe, S. M. *J. Am. Chem. Soc.* **1992**, *114*, 5555-5566.
55. Veretenov, A. L.; Koltun, D. O.; Smit, W. A.; Strelenko, Y. A. *Tetrahedron Lett.* **1995**, *36*, 4651-4654.
56. (a) Hua, D. H. *J. Am. Chem. Soc.* **1986**, *108*, 3835-3837. (b) Hua, D. H.; Coulter, M. J.; Badejo, I. *Tetrahedron Lett.* **1987**, *28*, 5465-5468.
57. Billington, D. C.; Willison, D. *Tetrahedron Lett.* **1984**, *25*, 4041-4044.
58. Negri, J. T.; Morwick, T.; Doyon, J.; Wilson, P. D.; Hickey, E. R.; Paquette, L. A. *J. Am. Chem. Soc.* **1993**, *115*, 12189-12190.

59. Paquette, L. A.; Doyon, J. *J. Am. Chem. Soc.* **1995**, *117*, 6799-6800.
60. Morwick, T.; Doyon, J.; Paquette, L. A. *Tetrahedron Lett.* **1995**, *36*, 2369-2372.
61. Paquette, L. A.; Morwick, T. *J. Am. Chem. Soc.* **1995**, *117*, 1451-1452.
62. (a) Paquette, L. A.; Doyon, J.; Kuo, L. H. *Tetrahedron Lett.* **1996**, *37*, 3299-3302. (b) Morwick, T. M.; Paquette, L. A. *J. Org. Chem.* **1996**, *61*, 146-152
63. (a) Santora, V. J.; Moore, H. W. *J. Am. Chem. Soc.* **1995**, *117*, 8486-8487. (b) MacDougall, J. M.; Moore, H. W. *J. Org. Chem.* **1997**, *62*, 4554-4555.
64. Jones, G. *Organic Photochemistry*; Padwa, A.; Ed.; Marcel-Dekker: New York, **1971**, 1-122.
65. (a) Rawal, V. H.; Dufour, C. *J. Am. Chem. Soc.* **1994**, *116*, 2613-2614. (b) Rawal, V. H.; Dufour, C.; Iwasa, S. *Tetrahedron Lett.* **1995**, *36*, 19-22. (c) Rawal, V. H.; Dufour, C.; Eschbach, A. *J. Chem. Soc., Chem. Commun.* **1994**, 1797-1798.
66. (a) Rawal, V. H.; Fabre, A.; Iwasa, S. *Tetrahedron Lett.* **1995**, *36*, 6851-6854. (b) Sternbach, D. D.; Ensinger, L. L. *J. Org. Chem.* **1990**, *55*, 2725-2736.
67. Meyer, C.; Marek, I.; Courtemanche, G.; Normant, J. F. *J. Org. Chem.* **1995**, *60*, 863-871.
68. Meyer, C.; Marek, I.; Normant, J. F. *Tetrahedron Lett.* **1996**, *37*, 857-860.
69. (a) Fournet, G.; Balme, G.; Gore, J. *Tetrahedron Lett.* **1989**, *30*, 69-70. (b) Balme, G.; Bouyssi, D. *Tetrahedron* **1994**, *50*, 403-414. (c) Vittoz, P.; Bouyssi, D.; Traversa, C.; Gore, J.; Balme, G. *Tetrahedron Lett.* **1994**, *35*, 1871-1874.
70. Ramig, K.; Kuzemko, M. A.; McNamara, K.; Cohen, T. D. *J. Org. Chem.* **1992**, *57*, 1968-1969.
71. (a) Cohen, T.; McNamara, K.; Kuzemko, M. A.; Ramig, K.; Landi, Jr. J. J.; Dong, Y.; *Tetrahedron* **1993**, *49*, 7931-7942. (b) Cohen, T.; Yu, L. C. *J. Org. Chem.* **1985**, *50*, 3266-3269. (c) Ramig, K.; Bhupathy, M.; Cohen, T. *J. Org. Chem.* **1989**, *54*, 4404-4412. (d) Smith, R. A. J.; Lal, A. R. *Aust. J. Chem.* **1979**, *32*, 353-360.
72. Ramig, K.; Bhupathy, M.; Cohen, T. D. *J. Am. Chem. Soc.* **1988**, *110*, 2678-80.
73. Trost, B. M.; Shi, Y. *J. Am. Chem. Soc.* **1993**, *115*, 9421-9438.
74. (a) Singh, V. K.; Deota, P. T.; Bedekar, A. *J. Chem. Soc., Perkin Trans. 1* **1992**, 903-912. (b) Singh, V. K.; Deota, P. T.; Raju, B. N. S. *Synth. Commun.* **1987**, *17*, 115-124. (c) Singh, V. K.; Raju, B. N. S.; Deota, P. T. *Indian J. Chem.* **1987**, *26B*, 301-304. (d) Singh, V. K.; Deota, P. T. *J. Chem. Res.* **1989**, *12*, 390. (e) Singh, V. K.; Raju, B. N. S. *Synth. Commun.* **1989**, *18*, 1513-1518. (f) Singh, V. K.; Bedekar, A. V. *Synth. Commun.* **1989**, *19*, 107-118. (g) Singh, V. K.; Porinchu, M.; *Tetrahedron* **1996**, *52*, 7087-7126.
75. Singh, V. K.; Prathap, S.; Porinchu, M. *Tetrahedron Lett.* **1997**, *38*, 2911-2914.
76. (a) Singh, V. K.; Thomas, B. *J. Chem. Soc., Chem. Commun.* **1992**, 1211-1213. (b) Singh, V. K.; Thomas, B. *J. Org. Chem.* **1997**, *62*, 5310-5320.
77. (a) Sternbach, D. D.; Ensinger, C. L. *J. Org. Chem.* **1990**, *55*, 2725-2736. (b) Sternbach, D. D. *Strategies and Tactics in Organic Synthesis*; Lindberg, T.; Ed.; Academic Press, Inc.

- SanDiego, California, 1989, Vol. 2, Chapter 12, 415-438. (d) Sternbach, D. D.; Hughes, J. W.; Burdi, D. F.; Banks, B. A. *J. Am. Chem. Soc.* 1985, 107, 2149-2153. (e) Sternbach, D. D.; Hughes, J. W.; Burdi, D. F.; Forstot, R. M. *Tetrahedron Lett.* 1983, 24, 3295-3298.
78. (a) Toyota, M.; Nishikawa, Y.; Motoki, K.; Yoshida, N.; Fukumoto, K. *Tetrahedron* 1993, 49, 11189-11204. (b) Toyota, M.; Nishikawa, Y.; Motoki, K.; Yoshida, N.; Fukumoto, K. *Tetrahedron Lett.* 1993, 34, 6099-6102.
79. (a) Mase, T.; Shibasaki, M.; *Tetrahedron Lett.* 1986, 27, 5245-5248. (b) Shibasaki, M.; Mase, T.; Ikegami, S. *J. Am. Chem. Soc.* 1986, 108, 2090-2091. (c) Kagechika, K.; Shibasaki, M. *J. Org. Chem.* 1991, 56, 4093-4094.
80. Kagechika, K.; Ohshima, T., Shibasaki, M. *Tetrahedron* 1993, 49, 1773-1782.
81. (a) Oppolzer, W.; Zurita, M. B.; Switzer, C. Y. *Tetrahedron Lett.* 1988, 29, 6433-6436. (b) Oppolzer, W.; Xu, J. Z.; Stone, C. *Helv. Chim. Acta* 1991, 74, 465-468. (c) Oppolzer, W.; Robyr, C. *Tetrahedron* 1994, 50, 415-424.
82. Asaoka, M.; Obuchi, K.; Takei, H. *Tetrahedron* 1994, 50, 655-660.
83. Piers, E; Karunaratne, V. *Tetrahedron* 1989, 45, 1089-1104.
84. Clive, D. L. J.; Magnuson, S. R. *Tetrahedron Lett.* 1995, 36, 15-18.

Biographical sketch



Vishwakarma Singh



Beena Thomas

Vishwakarma Singh received his Ph.D. degree in 1976 from University of Gorakhpur where he worked under the supervision of Prof. S. Giri. He continued his scientific education as a Senior Research Fellow (CSIR, New Delhi) under the direction of Prof. Goverdhan Mehta at Indian Institute of Technology, Kanpur and University of Hyderabad, where he worked in the area of strained polycyclics. He was awarded a Monbusho Fellowship from Ministry of Education, Govt. of Japan, to work with Prof. E. Osawa at Hokkaido University, Sapporo, Japan (1979-80). He then moved to United States and worked as a Research Associate with Prof. A. R. Martin and Dr. V. V. Kane at University of Arizona, Tucson, in the area of cannabinoids. Vishwakarma Singh also worked with Prof. J. B. Hendrickson at Brandeis University, Waltham, Mass, on synthesis of corticosteroids (1981-82).

Dr. Vishwakarma Singh returned to India and after working at Malti-Chem Research Center Nandesari, Baroda and M. S. University of Baroda, he joined Indian Institute of Technology, Bombay in October 1989 and currently he is an Associate Professor. His research interests are in the area of organic synthesis and photochemistry. He has completed total synthesis of triquinane natural products, $\Delta^{9(12)}$ -capnellene and coriolin and, currently he is working on synthesis of Taxol and Phorbol esters, photochemical reactions and asymmetric synthesis.

Beena Thomas completed her M.Sc. degree at Indian Institute of Technology, Bombay in 1990. Beena Thomas was awarded a research fellowship (CSIR, New Delhi) to continue with Doctoral studies at I.I.T. Bombay. She worked for her Ph.D. with Prof. Vishwakarma Singh in the field of tricyclopentanoid chemistry and received her Ph. D. degree in 1997. Beena has accepted a post-doctoral position at Ohio State University to work with Professor Y. H. Chu from 1st December, 1997.